



Enjeux et défis du déploiement massif du stockage électrochimique de l'énergie : matériaux organiques et éco-conception

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▪ UMR 6502 CNRS & Nantes Université



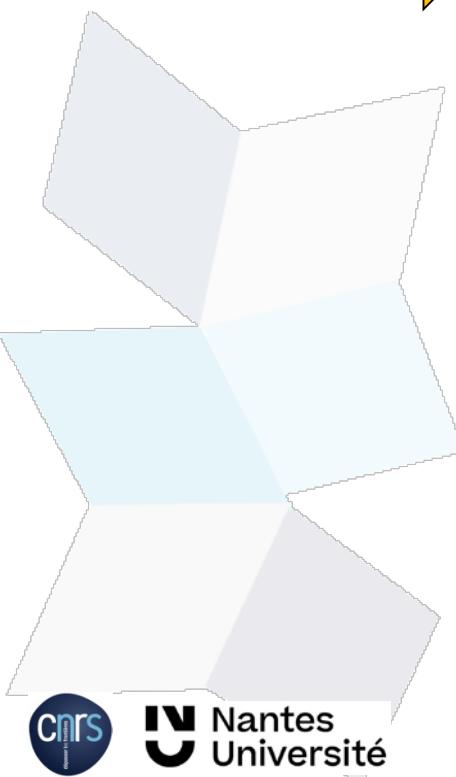
philippe.poizot@univ-nantes.fr

<https://www.cnrs-imn.fr/Philippe.Poizot>

PILE ≠ ACCUMULATEUR (... batterie d'accumulateurs)

Pile : difficulté à ramener l'état initial par simple recharge

- Système électrochimique non RECHARGEABLE
- Existe différents systèmes électrochimiques ; un exemple, les piles à combustible (O_2/H_2)
- Une pile rechargeable, ça n'existe pas !

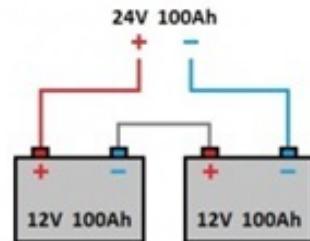


PILE ≠ ACCUMULATEUR (... batterie d'accumulateurs)

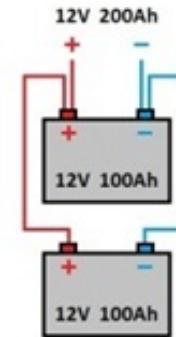
Accumulateur : les systèmes électrochimiques utilisés permettent le retour à l'état initial du système par recharge (en inversant le sens des réactions par inversion de la polarisation)



Batterie d'accumulateurs



BRANCHEMENT EN SÉRIE
Les tensions s'additionnent



BRANCHEMENT EN PARALLELE
Les intensités s'additionnent



Existe différents systèmes électrochimiques pour accumulateur
(Plomb-acide, Ni-Cd, Ni-MH, Li-ion)

Li-ion >2000 cycles de charge/décharge

Plan de l'exposé

I/ Situation mondiale en quelques chiffres

⇒ Données/sources robustes

II/ Vers une électrification massive de nos sociétés

⇒ Problématique du stockage de l'électricité, des batteries, des ressources...

III/ Batteries organiques

⇒ Une chimie complémentaire aux composés inorganiques

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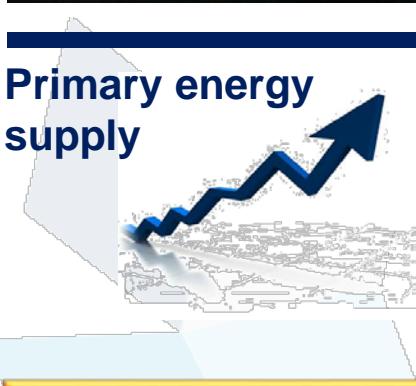
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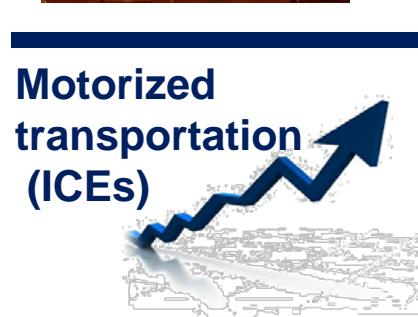
The key indicators of world: a constant increase!



Primary energy supply



Motorized transportation (ICEs)



Ore demand



Metal exhaust

GHG emissions



⇒ Booming from the industrial revolutions of the 19th century

⇒ Not sustainable

The key indicators of world: GHG & Global Warming

JOURNAL OF GEOPHYSICAL RESEARCH, VOL. 93, NO. D8, PAGES 9341–9364, AUGUST 20, 1988

Global Climate Changes as Forecast by Goddard Institute for Space Studies Three-Dimensional Model

J. HANSEN, I. FUNG, A. LACIS, D. RIND, S. LEBEDEFF, R. RUEDY, AND G. RUSSELL

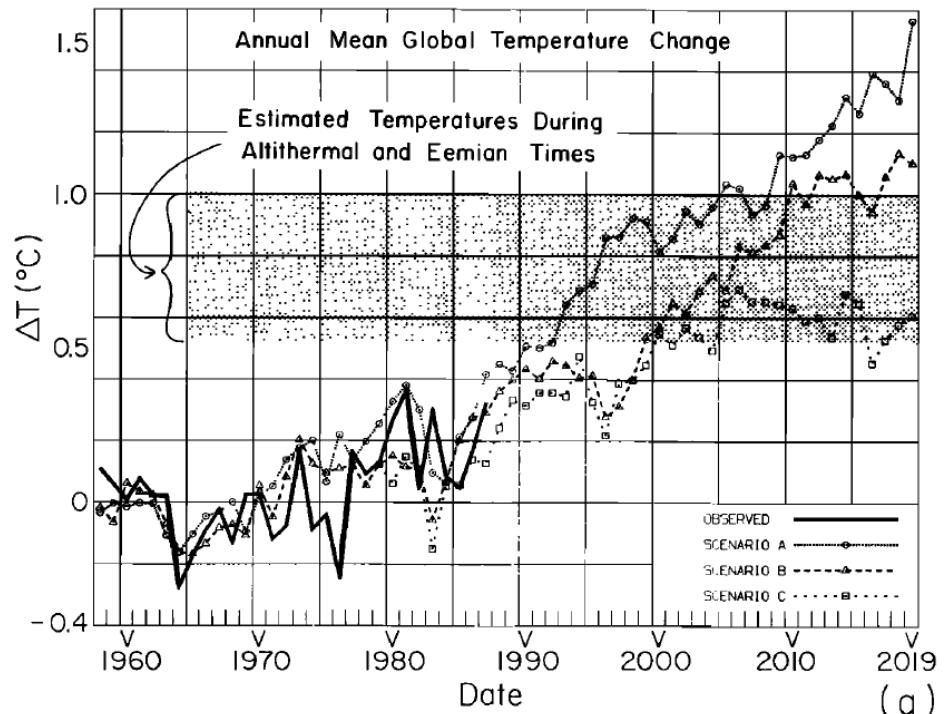
NASA Goddard Space Flight Center, Goddard Institute for Space Studies, New York

P. STONE
Massachusetts Institute of Technology, Cambridge

1988 : création du Groupe d'experts
intergouvernemental sur l'évolution du climat
(GIEC ou IPCC en anglais)

1987: Sustainable development: concept of the
public interest was proposed by the Brundtland
Commission

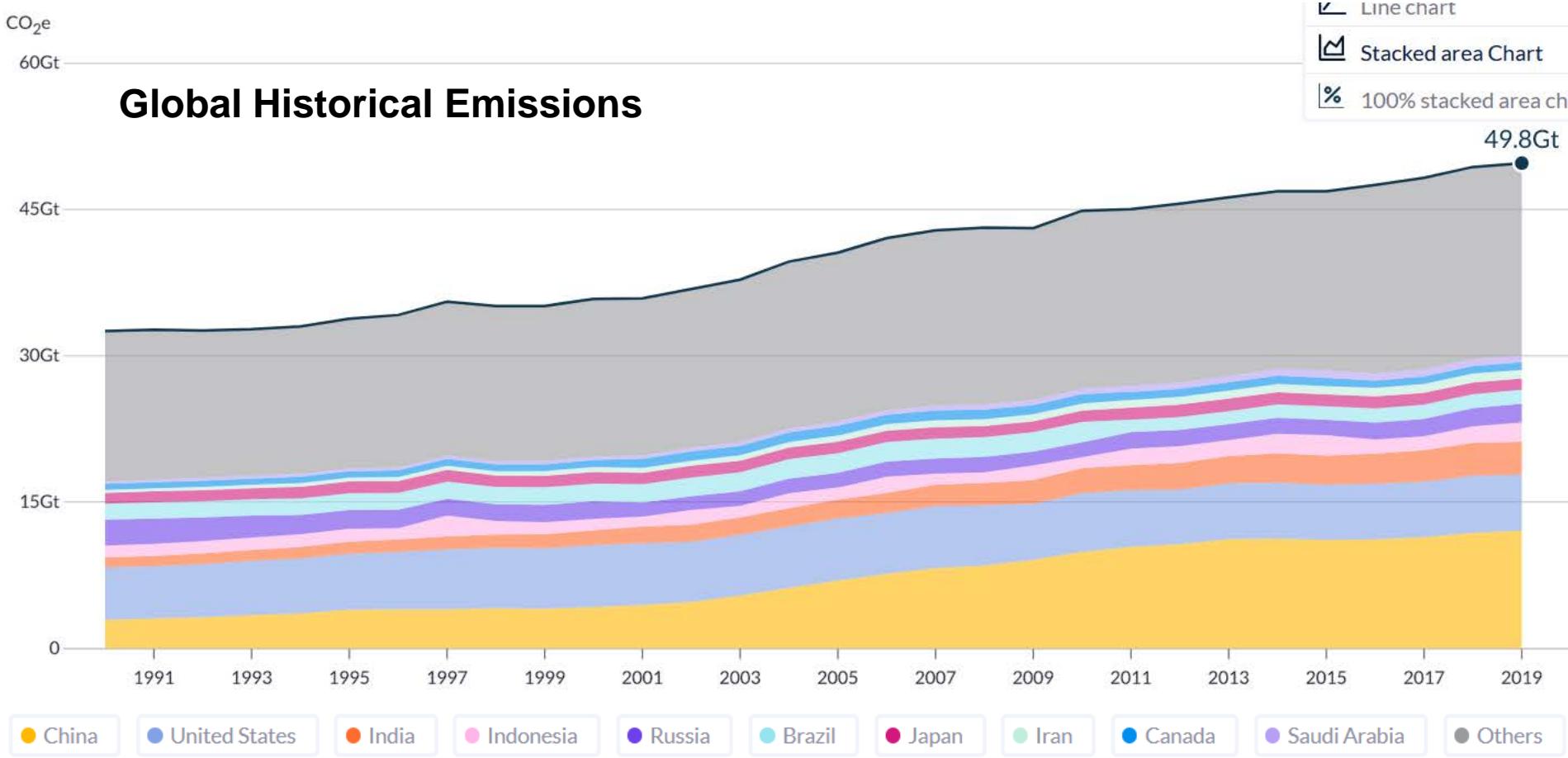
⇒ G. Brundtland, **Our Common Future: The**
World Commission on Environment and
Development, Oxford University Press, Oxford.



The key indicators of world: GHG & Global Warming

CLIMATEWATCH

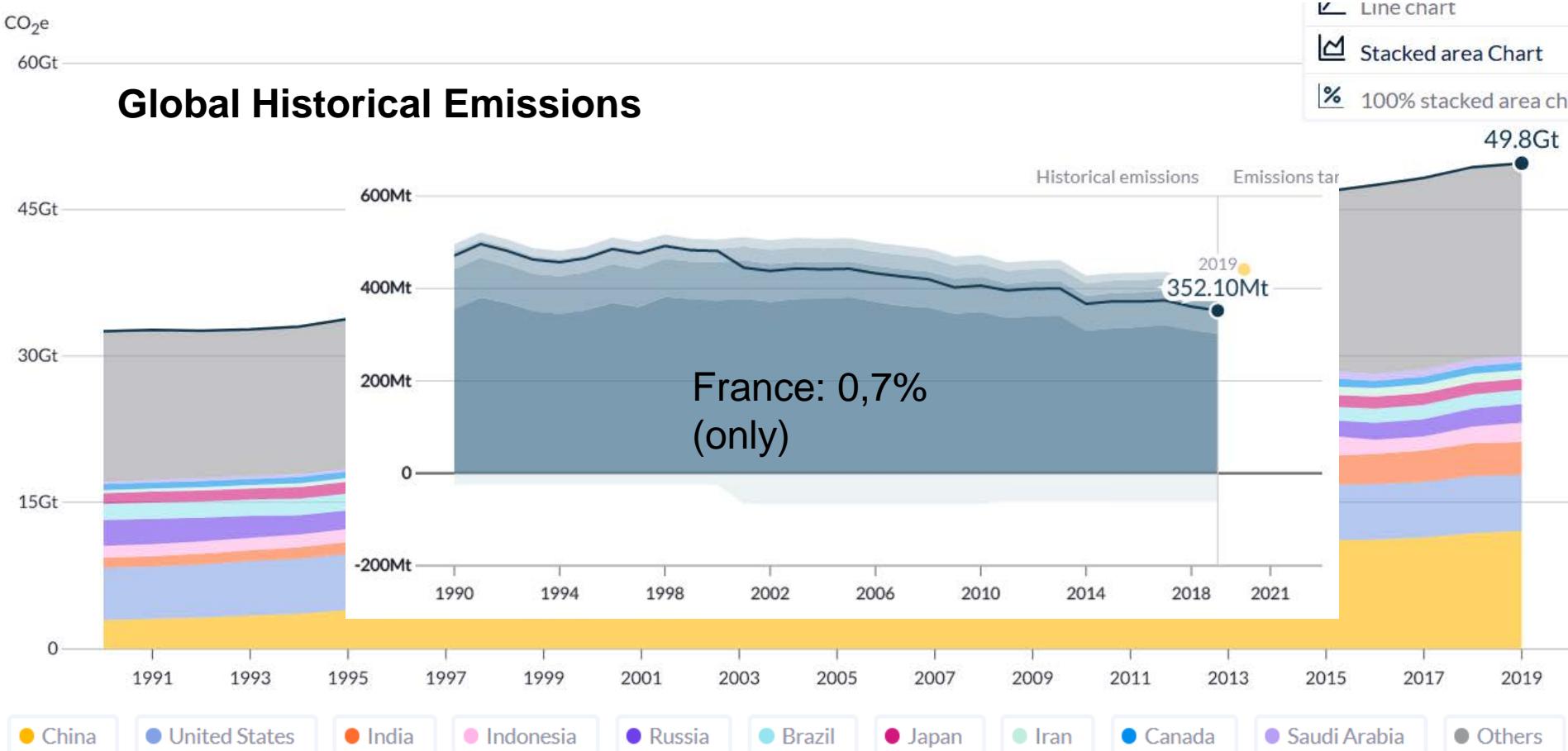
World Resources Institute (WRI) - <https://www.climatewatchdata.org/ghg-emissions>



The key indicators of world: GHG & Global Warming

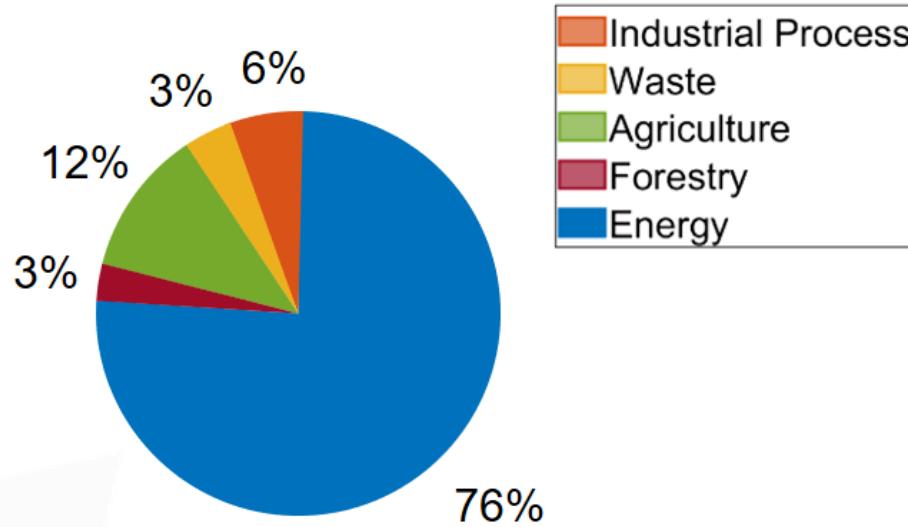
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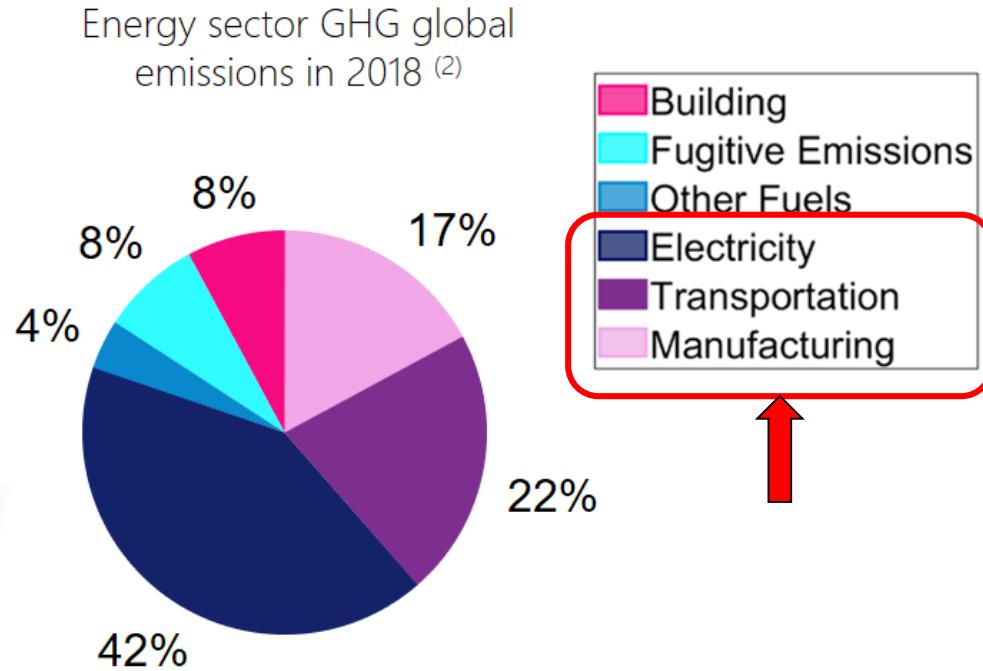
GHG global emissions in 2018 (2)



76 % of CO₂ emissions comes from the energy sector

(2) World Resources Institute (WRI) - <https://www.climatewatchdata.org/ghg-emissions>

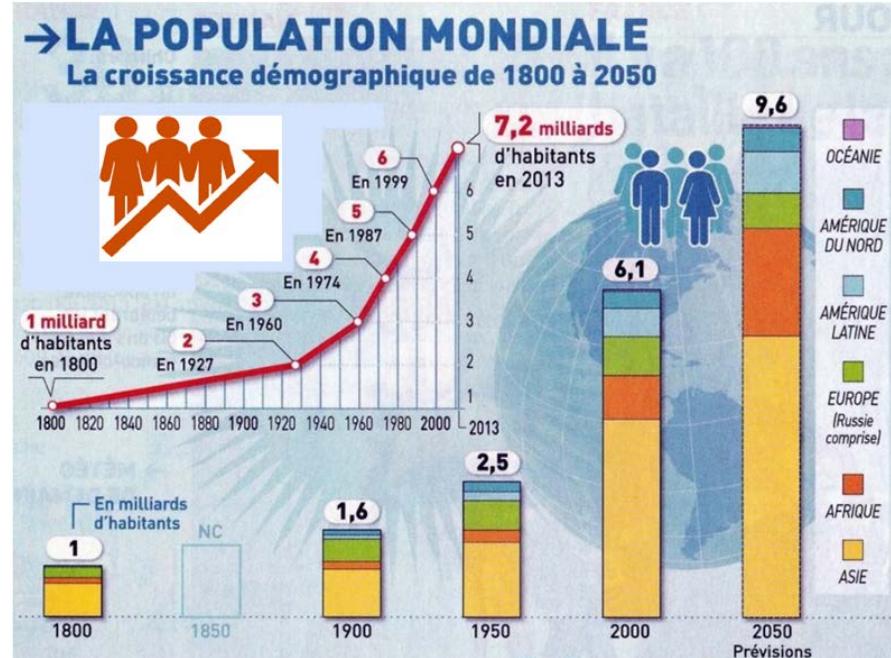
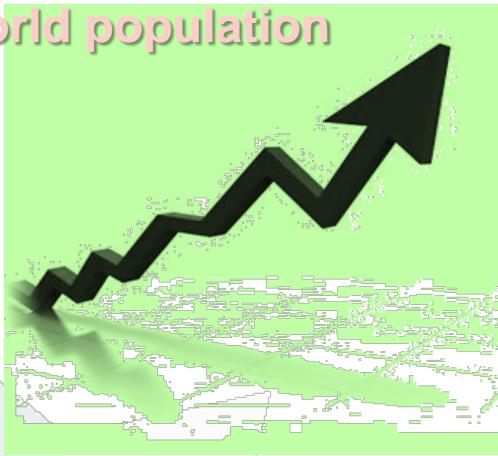
The key indicators of world: GHG & Global Warming



(2) World Resources Institute (WRI) - <https://www.climatewatchdata.org/ghg-emissions>

The key indicators of world: World population

World population



From UN data, <https://www.un.org/development/desa/>

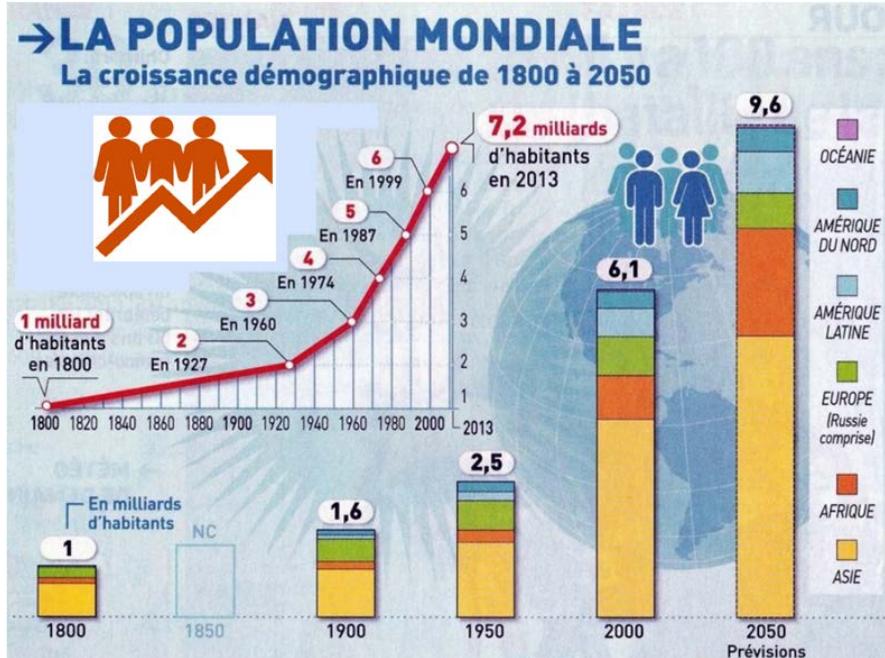
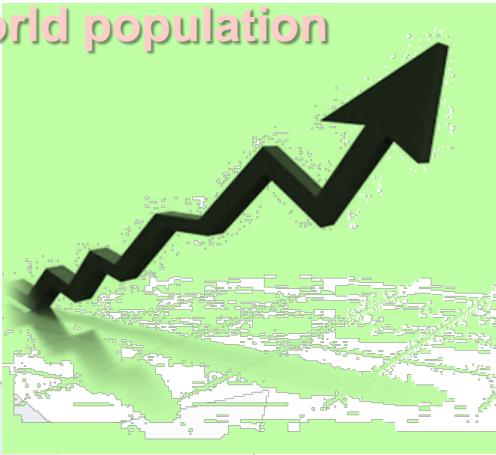
<https://www.futuribles.com/fr/revue/434/la-population-mondiale-a-lhorizon-2100-critique-de/>

⇒ 79% of the Human load localized in « emerging » countries

⇒ Human being = consumer especially in developed countries (GHGs makers)

The key indicators of world: World population

World population



"Human demand may well have exceeded the biosphere's regenerative capacity since the 80's"

(Wackernagel et al., PNAS, 2002, 99, 9266)

⇒ Changes in our daily life are needed

⇒ New population policies???

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⇒ Données/sources robustes

II/ Vers une électrification massive de nos sociétés

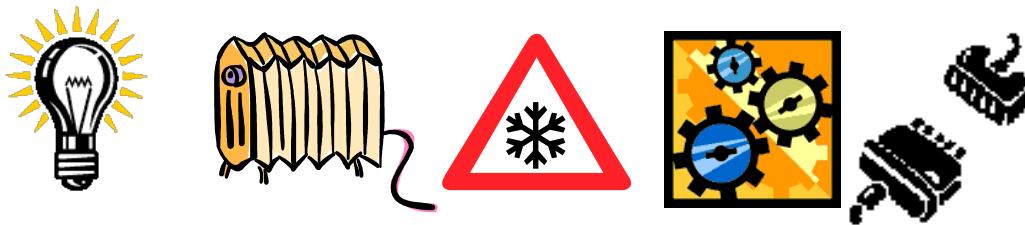
⇒ Problématique du stockage de l'électricité, des batteries, des ressources...

III/ Batteries organiques

⇒ Une chimie complémentaire aux composés inorganiques

Reconsider our energy engineering...

1 - Decarbonize the Power supply



Electricity = Functionality /everyday life necessity



LIFE EXPECTANCY

Rapid depletion when the electricity consumption per inhabitant is lower than 1600 kWh per year

Energy, 23, 791–801 (1998)

Reconsider our energy engineering...

1 - Decarbonize the Power supply



Electricity = Functionality /everyday life necessity

2 - Decarbonize the Transportation sector currently based on ICEs



A necessary trend: Electrification of the Transportation Systems

...Entering a so-called 4th Industrial Revolution

3 - Digitalization, IoT, service robots,... the growing consumption of electronic devices



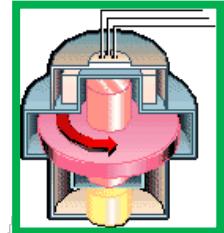
<https://martechtoday.com/>, Accessed June 2017

□ *Chem. Rev.*, **120**, 6490–6557 (2020)

The critical electricity storage issue: Only a few technological solutions



Capacitors
Super capacitors



Flywheels



Dam
pumping
station

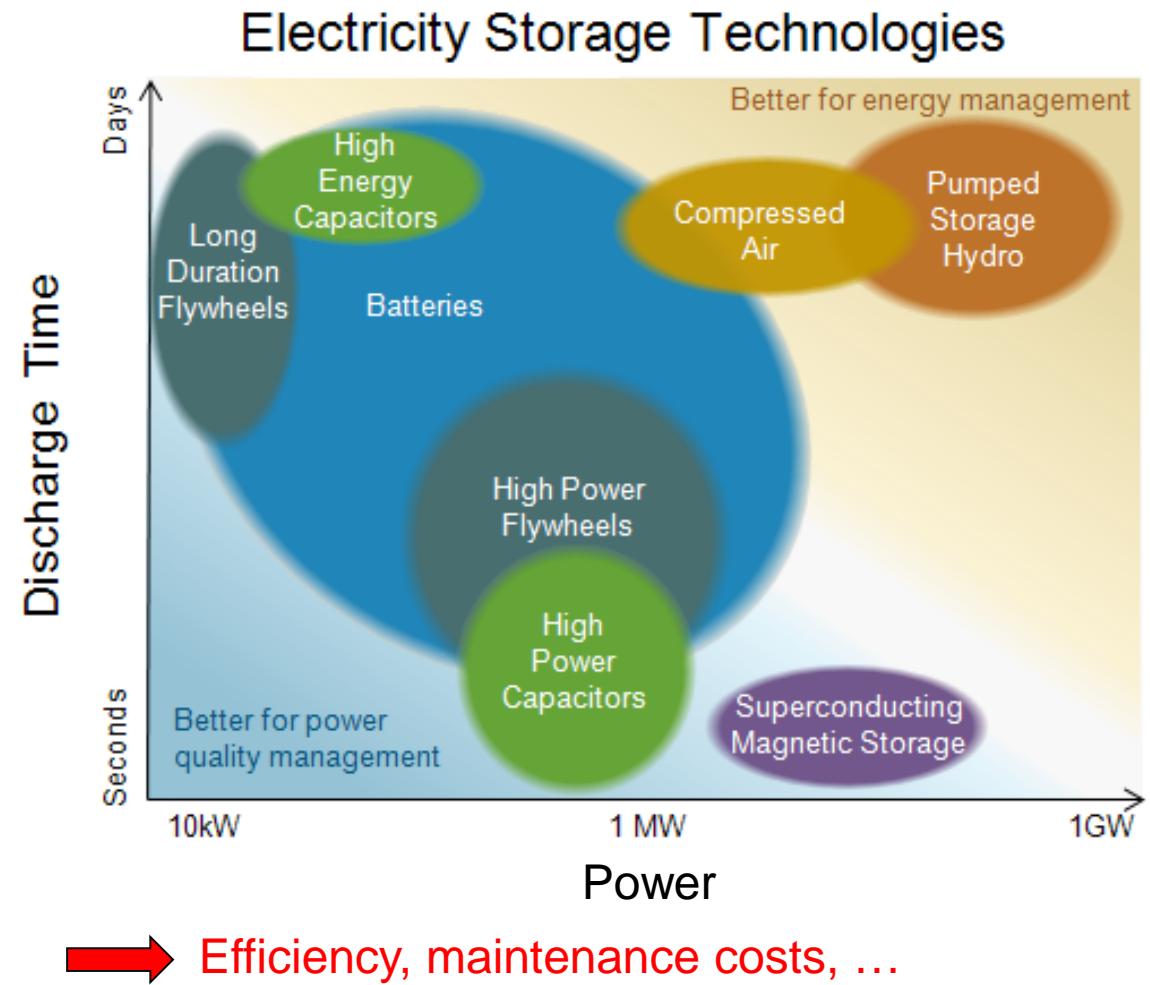
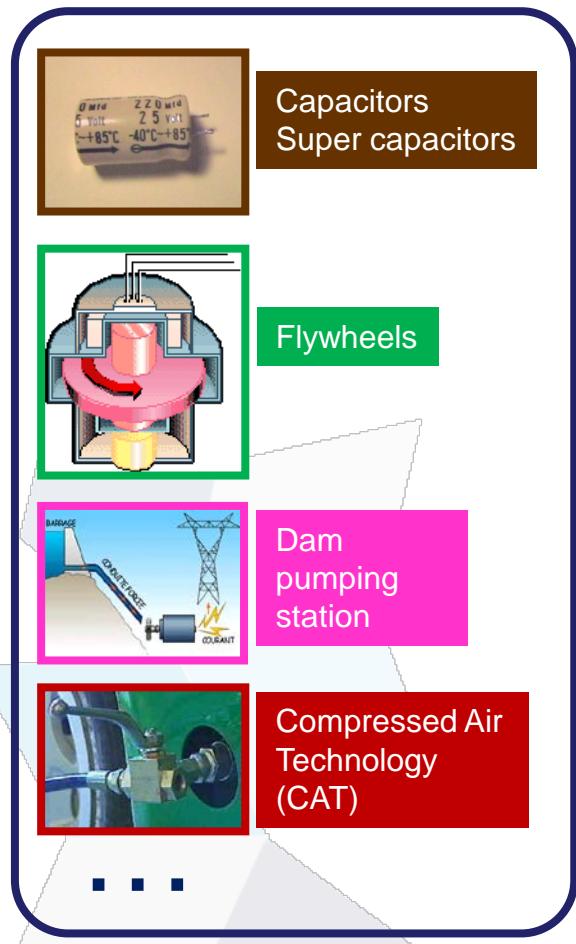


Compressed Air
Technology
(CAT)

...

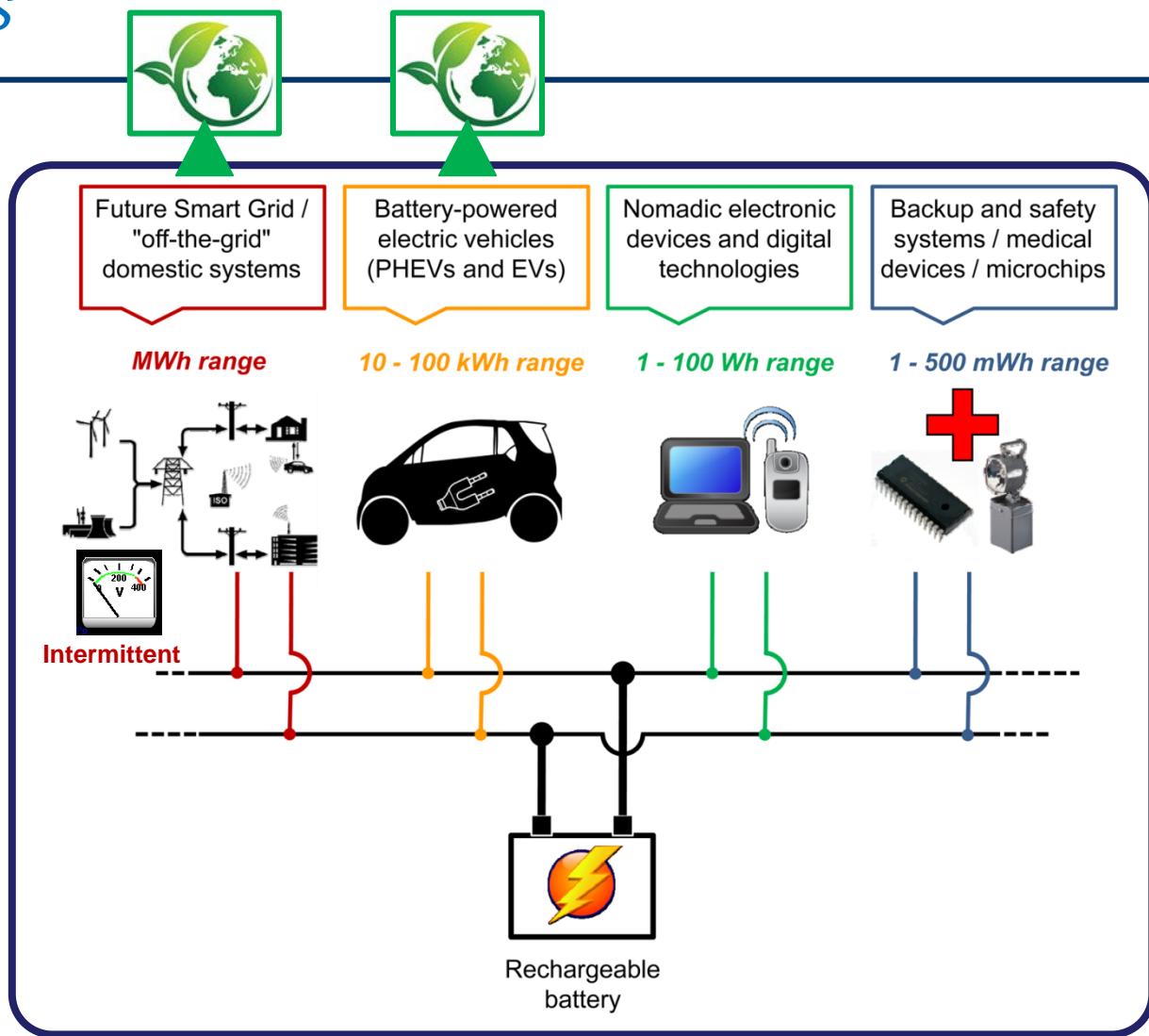
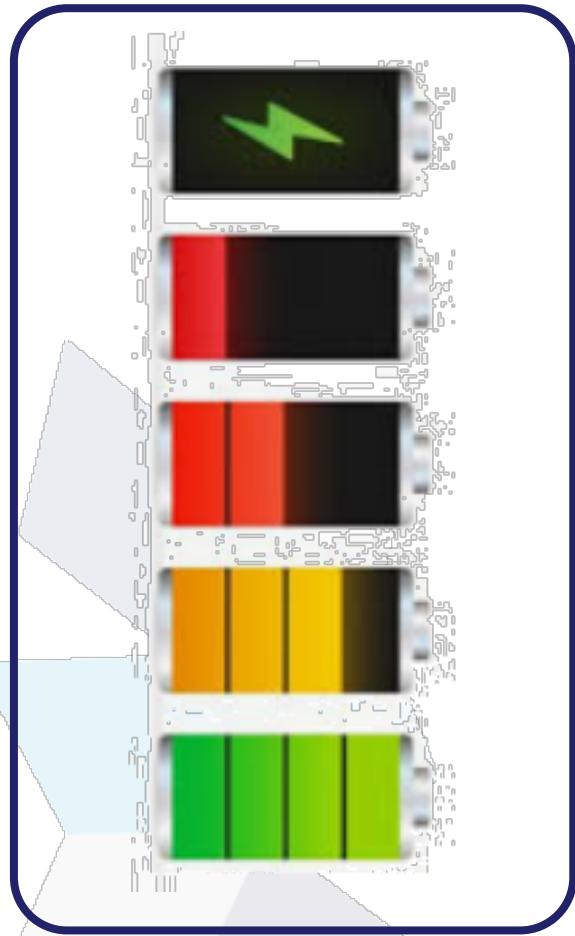
The critical electricity storage issue:

Only a few technological solutions

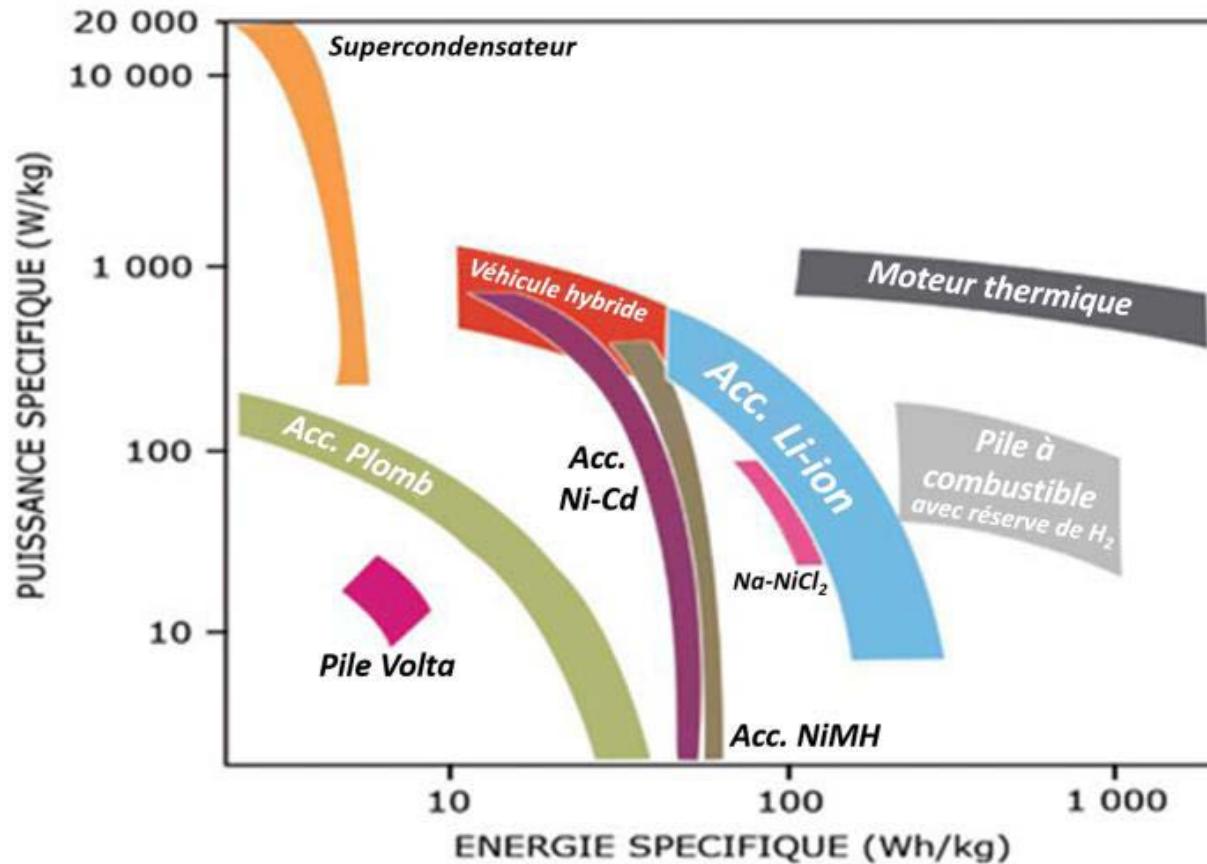


The critical electricity storage issue:

Rechargeable BATTERIES



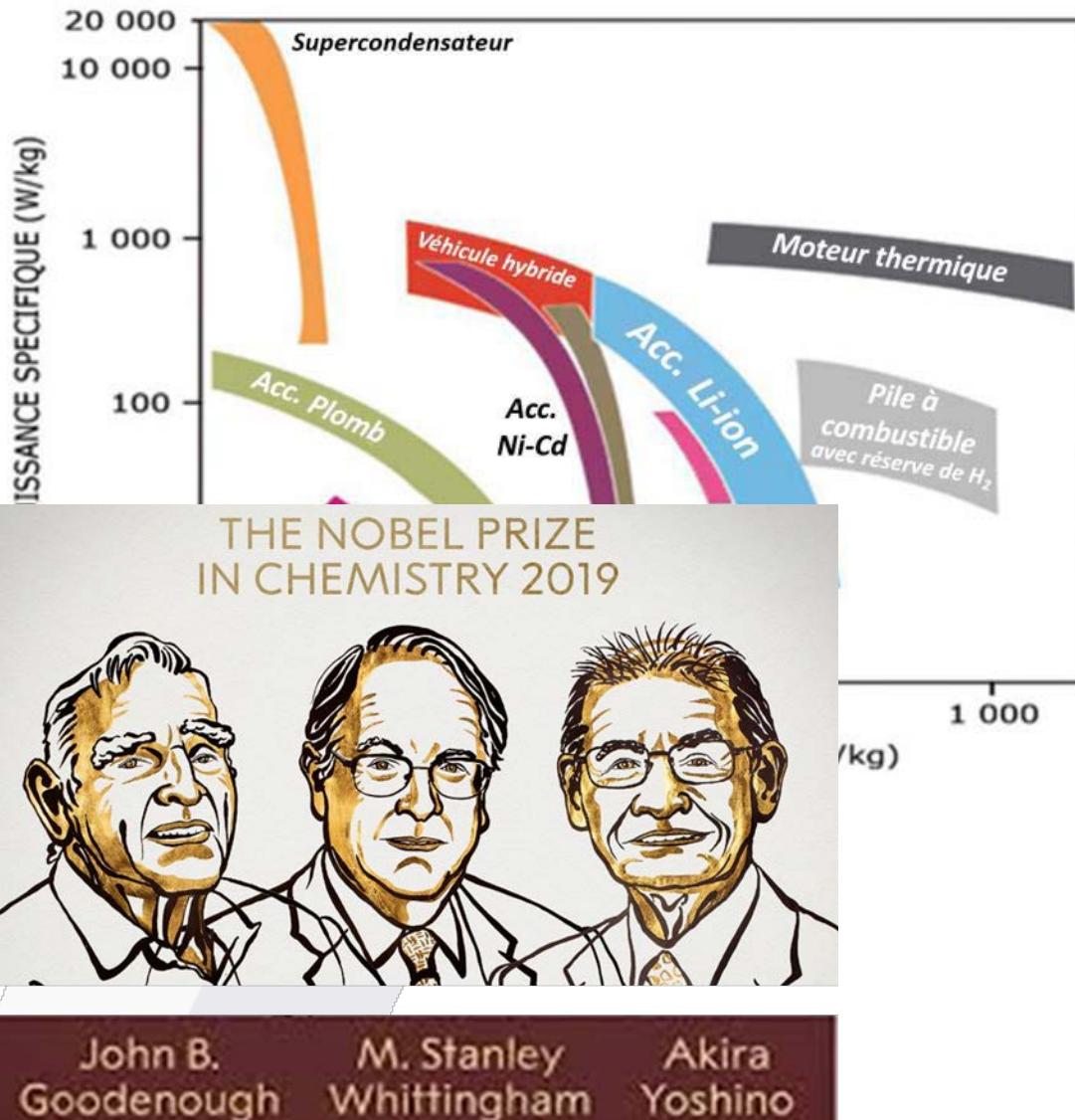
The critical electricity storage issue: Li-ion batteries, the flagship technology



Le Li-ion est devenu la technologie universelle pour l'électronique et les véhicules électrifiés (mais pas encore pour le stationnaire = batteries Pb-acide)

**Efficacité Energétique
Li-ion > 85% !!
2000 cycles 10 ans**

The critical electricity storage issue: Li-ion batteries, the flagship technology

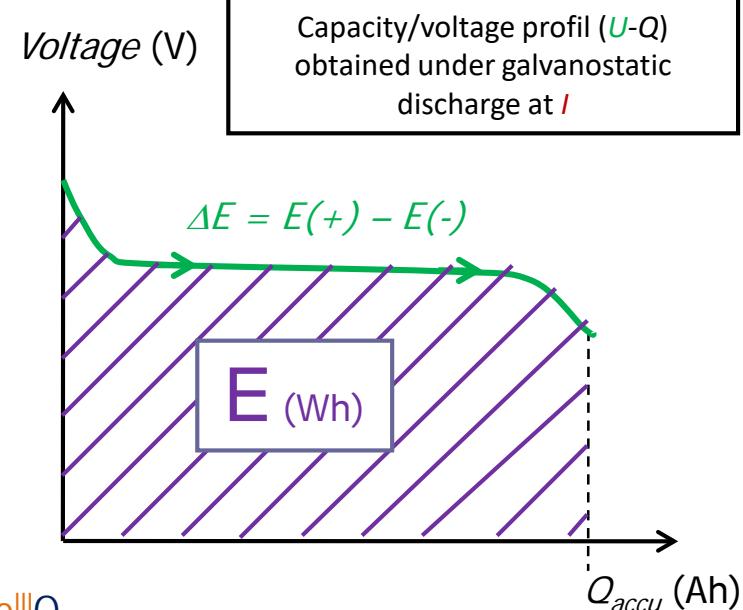
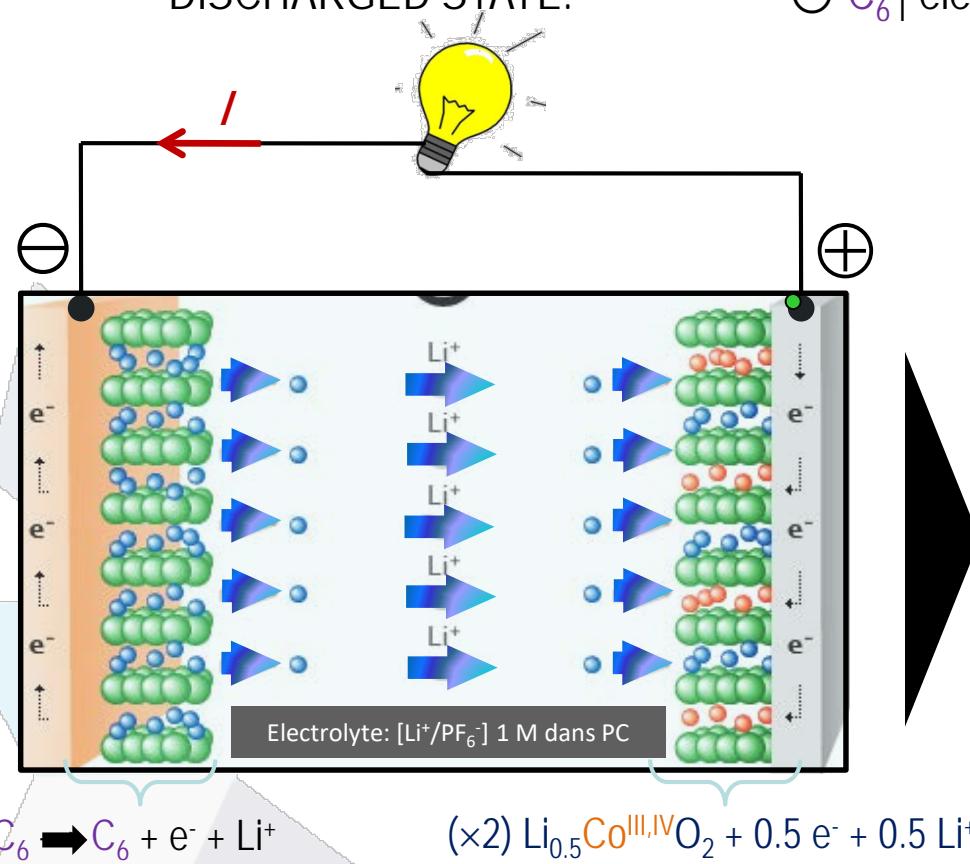


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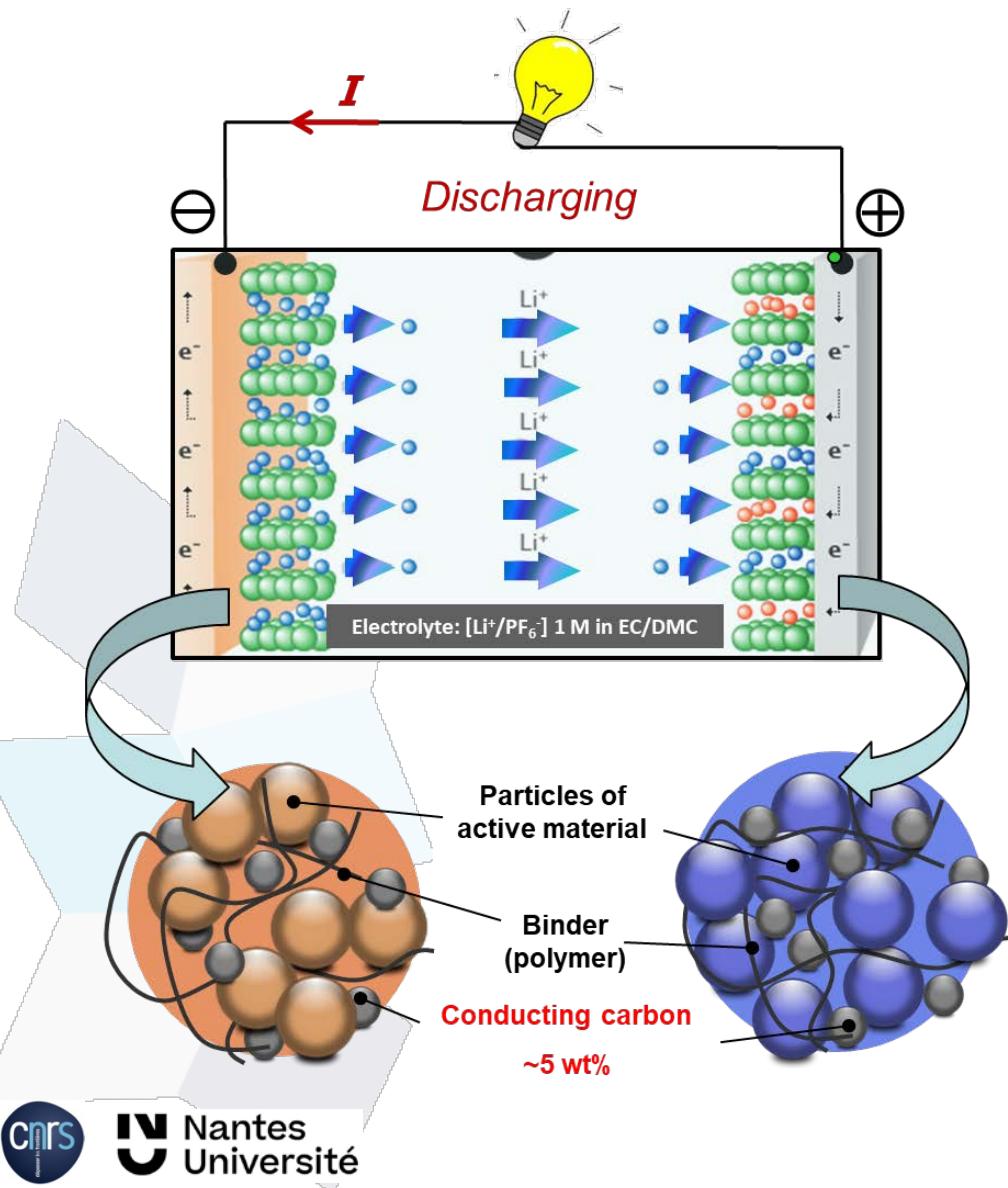
Working principle of a Li-ion cell: historical example (SONY, 1991)

CHARGED STATE (Initial state): $\ominus \text{LiC}_6 | \text{electrolyte} | \text{Li}_{0.5}\text{Co}^{\text{III,IV}}\text{O}_2 \oplus$
 DISCHARGED STATE: $\ominus \text{C}_6 | \text{electrolyte} | \text{LiCo}^{\text{III}}\text{O}_2 \oplus$

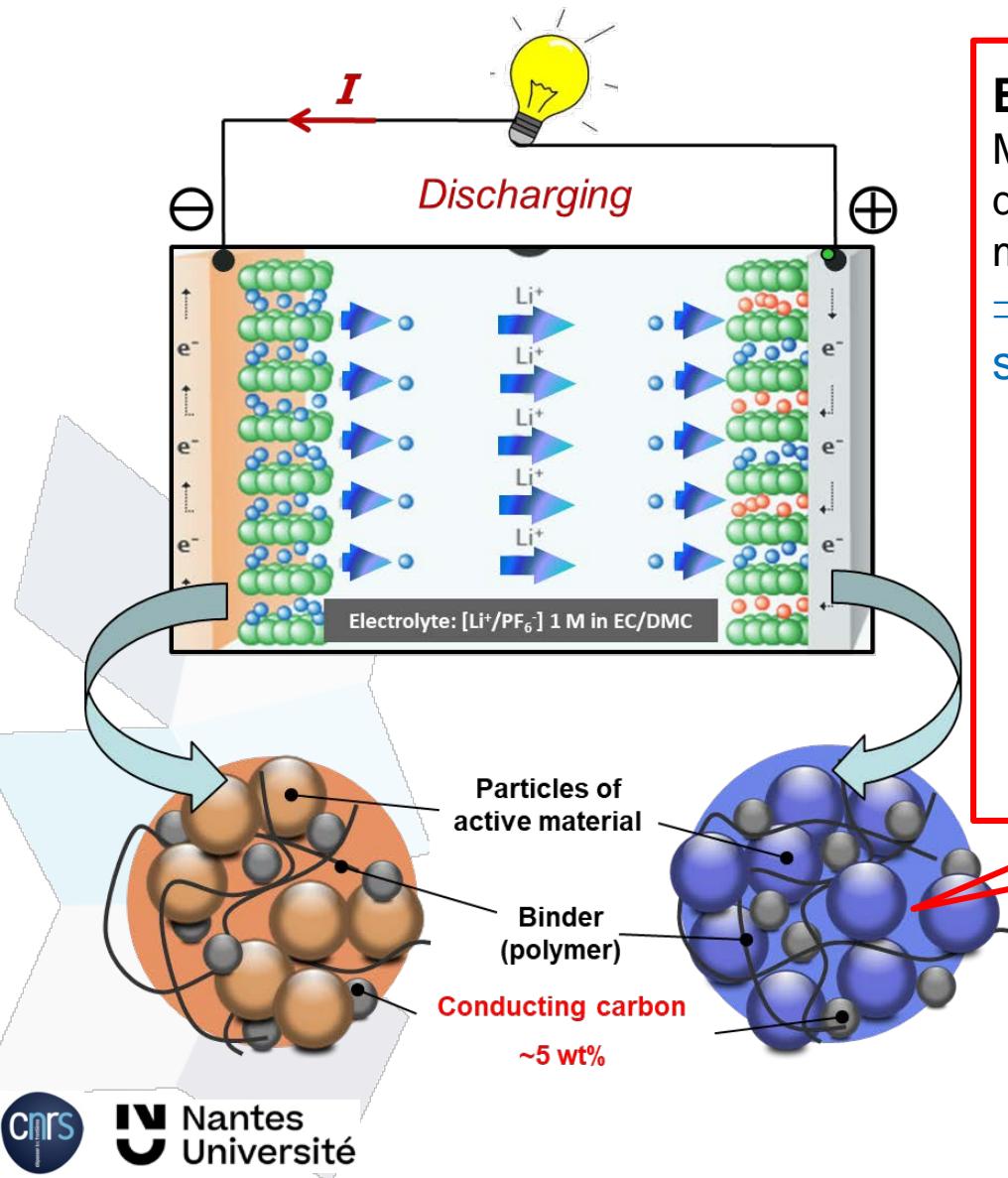


⇒ **Insertion materials as electroactive chemical compounds**

Working principle of a Li-ion cell: historical example (SONY, 1991)



Working principle of a Li-ion cell: historical example (SONY, 1991)



Encre d'électrode :

Matière (électro)active+liant+carbone+solvant du liant, déposé sur collecteur de courant métallique

⇒ Étape de séchage (évaporation du solvant)



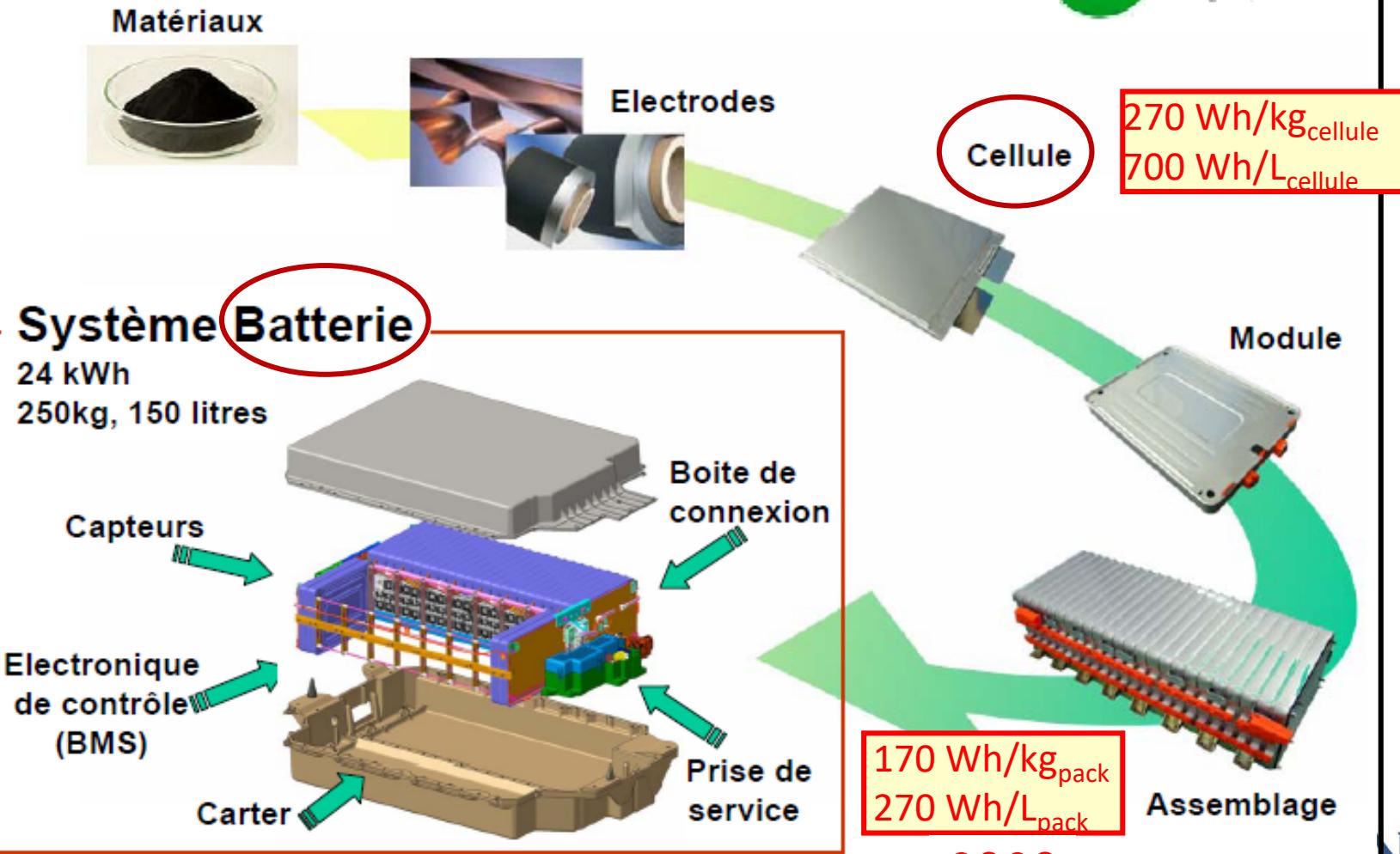
Working principle of a Li-ion cell: from the active materials to the battery pack

Fabrication et assemblage des batteries



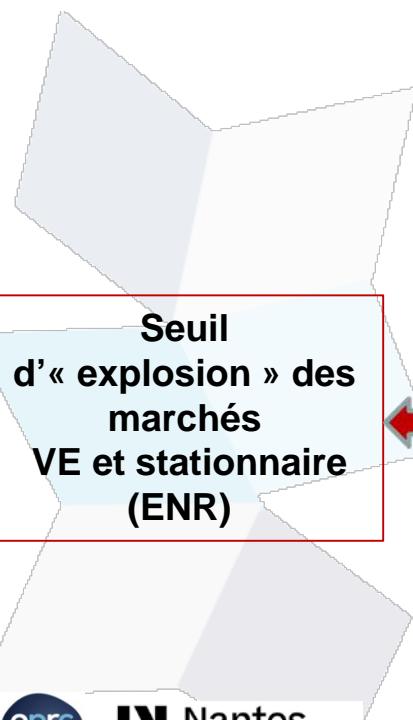
Automotive
Energy Supply
Corporation

2020



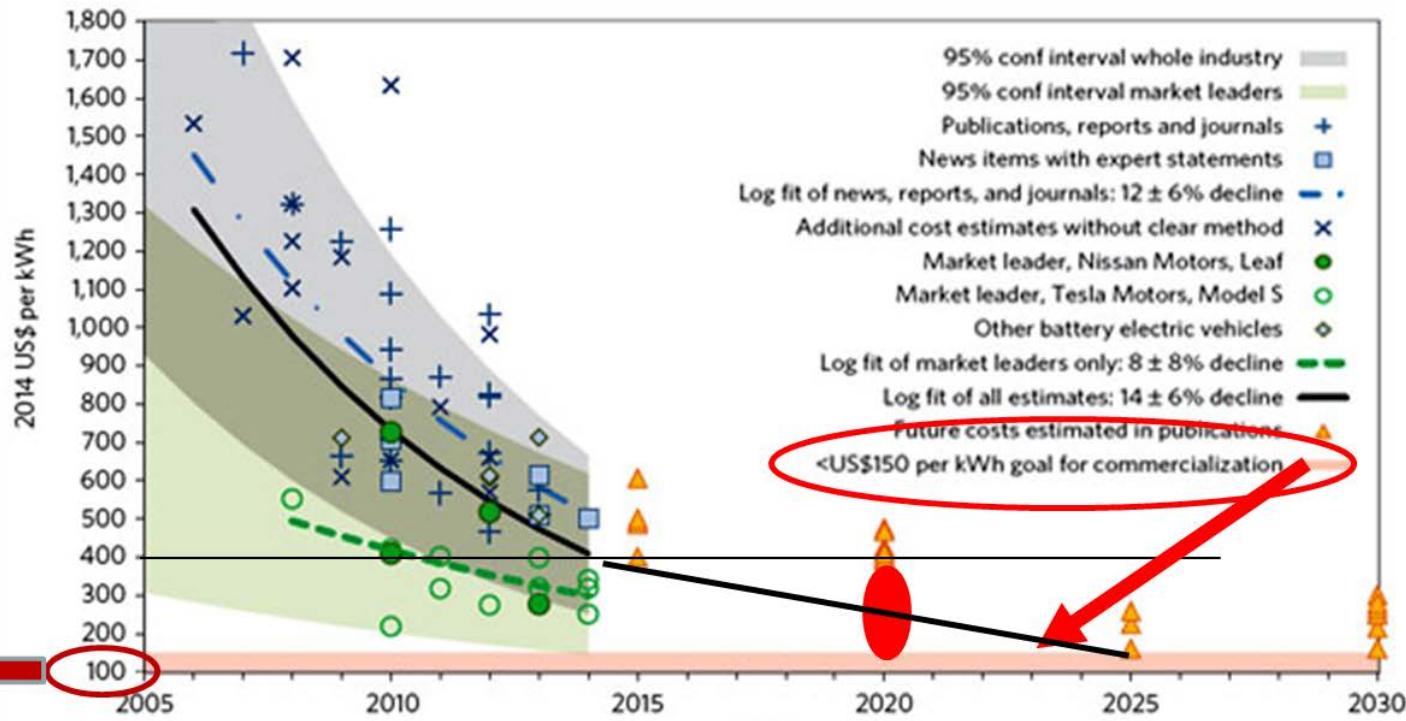
Cost of Li-ion batteries

Coût divisé par
5 en 10 ans
2020 ⇔ 150
\$/kWh

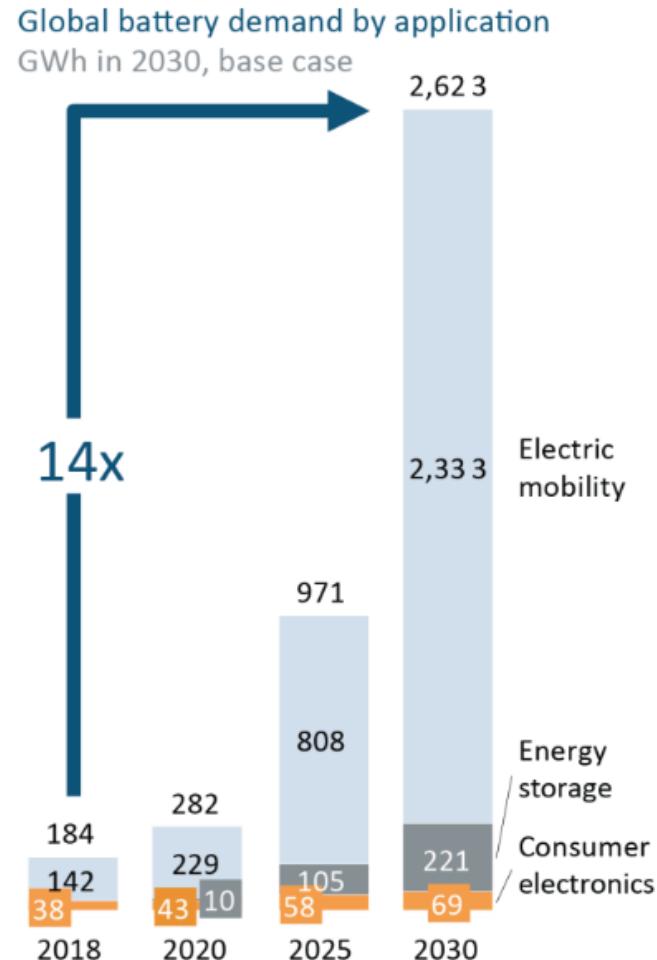


Seuil
d'« explosion » des
marchés
VE et stationnaire
(ENR)

Cost of Li-ion battery packs in battery electric vehicles



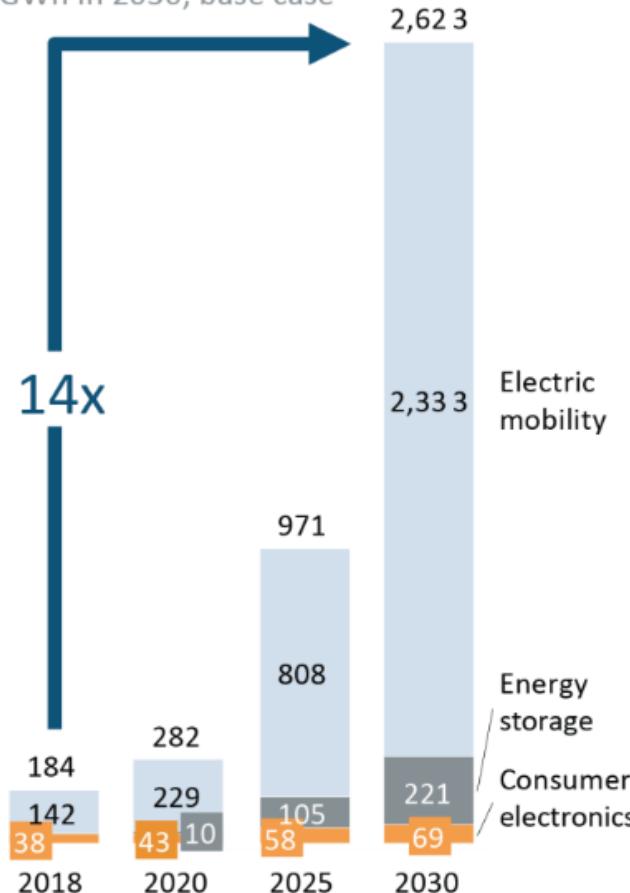
Big pressure on secondary batteries (especially LIBs) → *Global demand & projections*



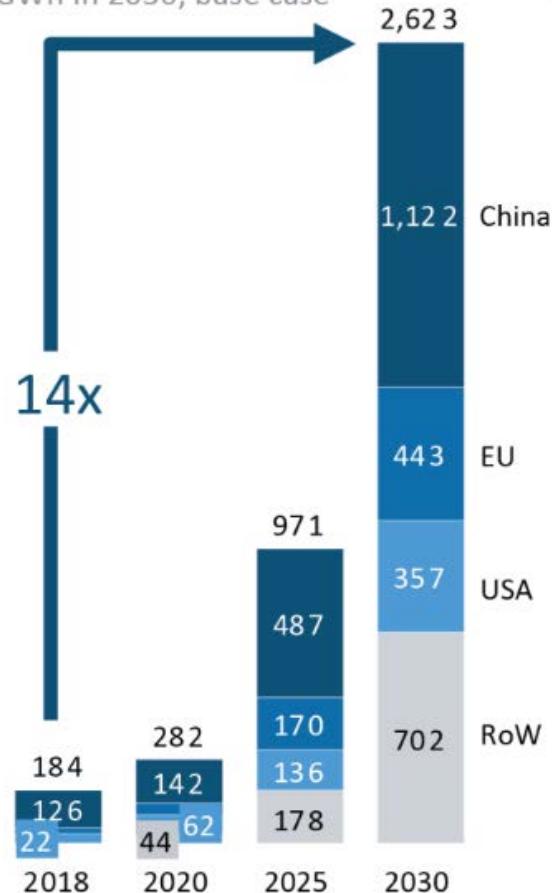
□ BATTERY 2030+ Roadmap

Big pressure on secondary batteries (especially LIBs) → *Global demand & projections*

Global battery demand by application
GWh in 2030, base case



Global battery demand by region
GWh in 2030, base case



Big pressure on secondary batteries (especially LIBs) → *Global demand & projections*



Gigafactory:

- Factory producing batteries at a very large scale
- Europe: 600 GWh/y full regime production

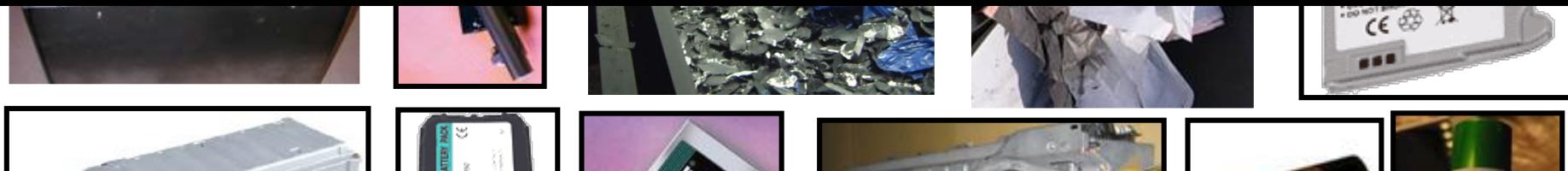


Common batteries: chemistry of metals and ceramic chemistry



Electrode materials (+) and (-):
Based on inorganic compounds (metal-based redox systems)

- synthesized from high temperature reactions
- obviously from non-renewable resources (polluting mining operations)
- energy-greedy recycling process (mostly pyro-metallurgy processes)
- battery cost depends on speculation in metals



On large scale development these drawbacks could be significant,
which can lessen the benefit of the present batteries

RECYCLABILITY: LCA studies for LIBs - A REVIEW

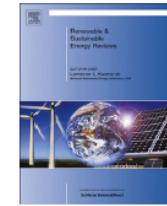
Renewable and Sustainable Energy Reviews 67 (2017) 491–506



Contents lists available at ScienceDirect

Renewable and Sustainable Energy Reviews

journal homepage: www.elsevier.com/locate/rser



The environmental impact of Li-Ion batteries and the role of key parameters – A review



Jens F. Peters ^{a,*}, Manuel Baumann ^{b,c}, Benedikt Zimmermann ^b, Jessica Braun ^b,
Marcel Weil ^{a,b}

- 1/ review of 113 available publications on the topic.
- 2/ a total of 36 LCA studies were identified as very reliable.
- 3/ **≈320 kWh** is needed across all chemistries to produce **1 kWh** of stored electrochemical energy producing GHG emissions of **110 kgCO₂eq**.
- 4/ “cradle-to-gate analyses” and **not “cradle-to-grave analyses”**, both collection and recycling steps are not included.
- 5/ other Eco indicators such as human toxicity (HTP) are not really studied and might be even more important (from mining excavation to the recycling step).

RECYCLABILITY & Eco-design: new regulations

DIRECTIVE 2006/66/EC OF THE EUROPEAN PARLIAMENT AND OF THE COUNCIL
of 6 September 2006
on batteries and accumulators and waste batteries and accumulators and repealing Directive
91/157/EEC
(Text with EEA relevance)

ANNEX III

DETAILED TREATMENT AND RECYCLING REQUIREMENTS

PART B: RECYCLING

3. Recycling processes shall achieve the following minimum recycling efficiencies:
 - (a) recycling of 65 % by average weight of lead-acid batteries and accumulators, including recycling of the lead content to the highest degree that is technically feasible while avoiding excessive costs;
 - (b) recycling of 75 % by average weight of nickel-cadmium batteries and accumulators, including recycling of the cadmium content to the highest degree that is technically feasible while avoiding excessive costs; and
 - (c) recycling of 50 % by average weight of other waste batteries and accumulators.

⇒ New revision of this Directive in progress with higher expectations

From: <https://eur-lex.europa.eu/legal-content/FR/ALL/?uri=CELEX%3A32018R2066>

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⇒ Données/sources robustes

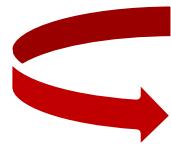
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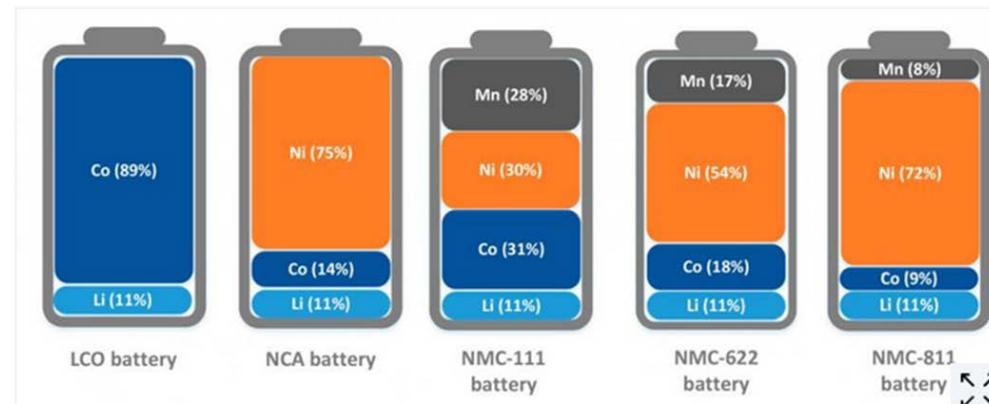
... promoting organic batteries?



Can be seen as a possible alternative to stabilize the pressure on redox-active metals vs. global demand

Cobalt as a component of electric vehicle batteries

We may tend to think that lithium is the major component of lithium-ion batteries (LIBs). However, cobalt generally makes up a greater percentage of the total. Various types of LIBs are currently used in electric vehicles (EVs), but the most important is the lithium nickel-manganese-cobalt oxide (NMC) type.



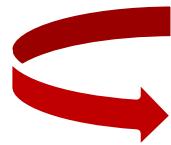
From:

<https://www.bgs.ac.uk/news/cobalt-resources-in-europe-and-the-potential-for-new-discoveries/>

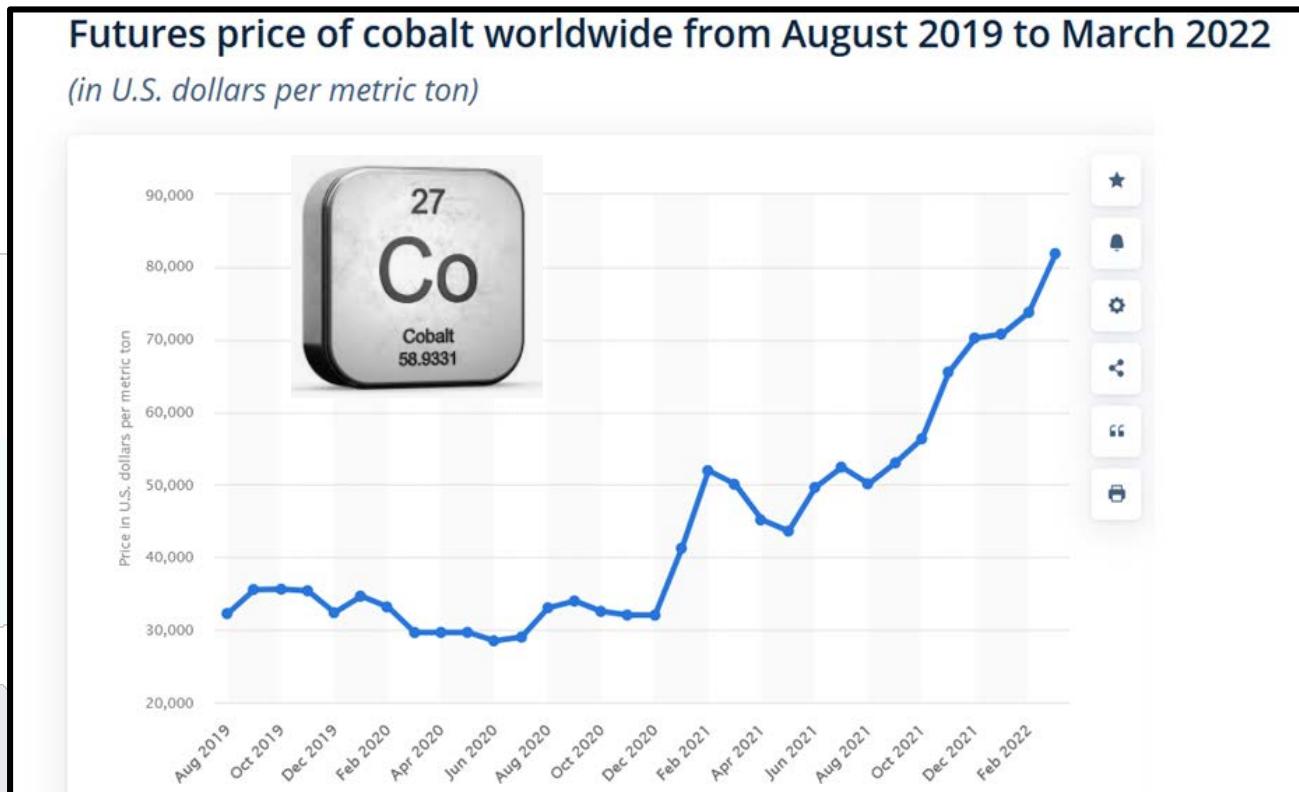


Lithium cobalt oxide (LCO) batteries, which contain about 89 per cent cobalt, are used in mobile phones, laptops and cameras. Lithium nickel-manganese-cobalt oxide (NMC) batteries, which typically contain between 9 and 31 per cent cobalt, are used in e-bikes and electric vehicles. Image: BGS © UKRI (based on figures by Olivetti et al., 2017).

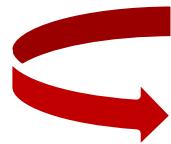
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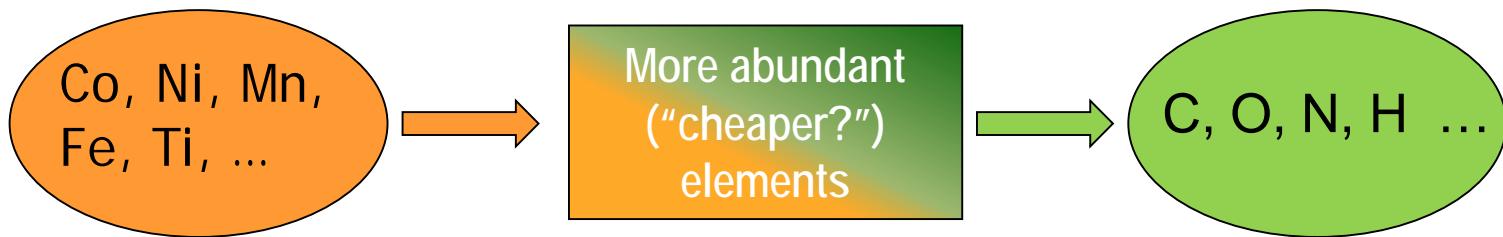
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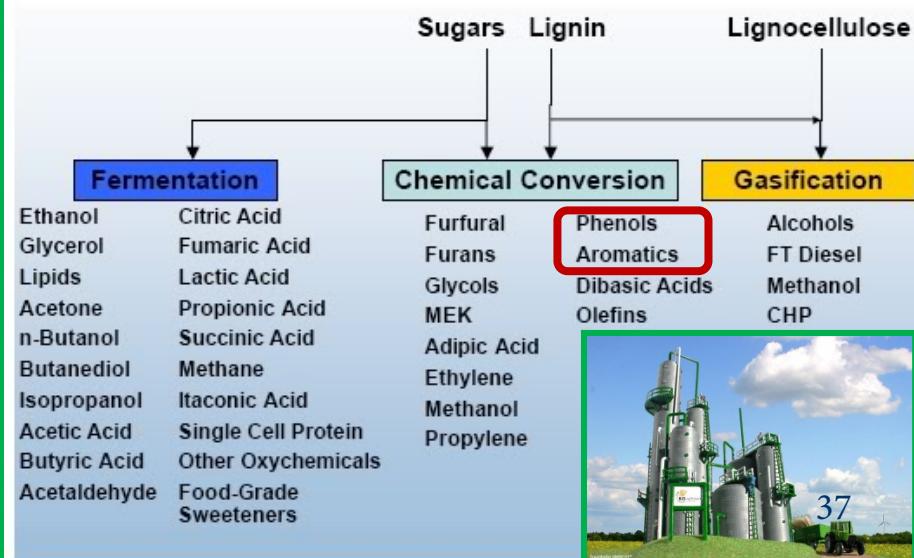
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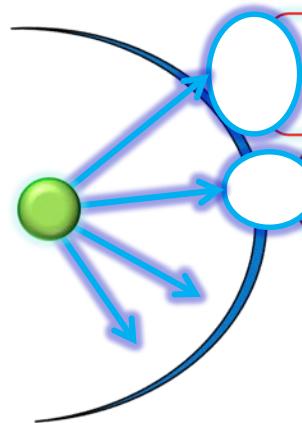
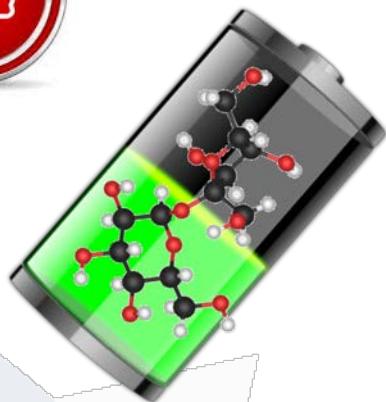
Can be seen as a possible alternative to stabilize the pressure on redox-active metals vs. global demand



Emergence of Biorefineries



Organic electrode materials, CONS



Low thermal stability
(but not systematically)

Lower volumetric energy density

- Typical volumetric mass density of inorganic host materials:

LiCoO_2 : 4.92 g/cm³

LiFePO_4 : 3.47 g/cm³

>

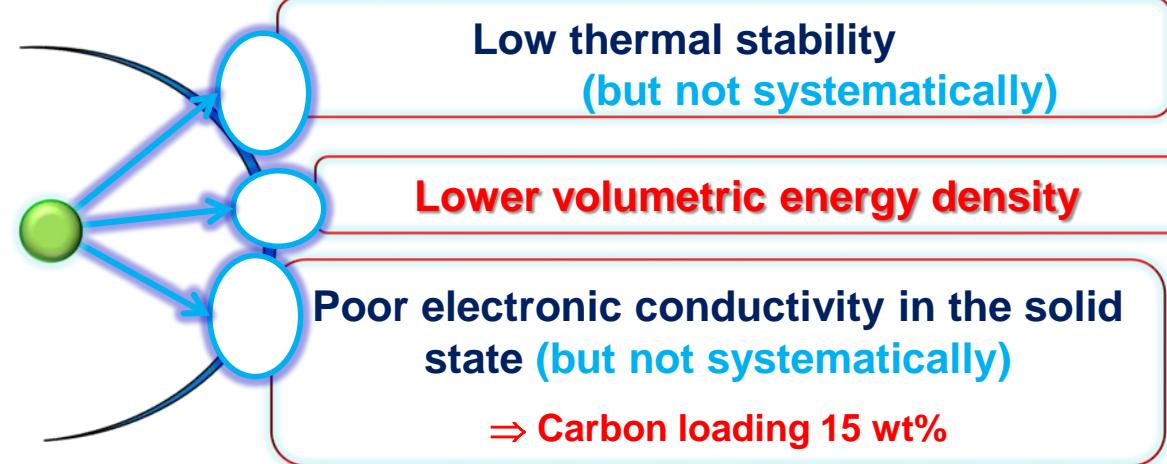
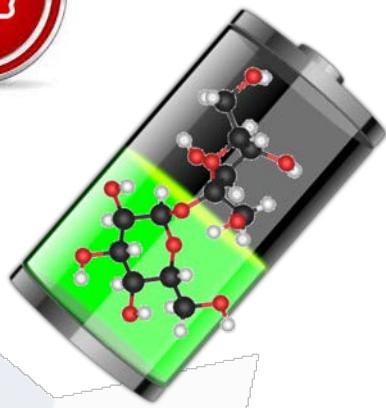
- Typical volumetric mass density of organique electrode materials:

1-2 g/cm³ (only)

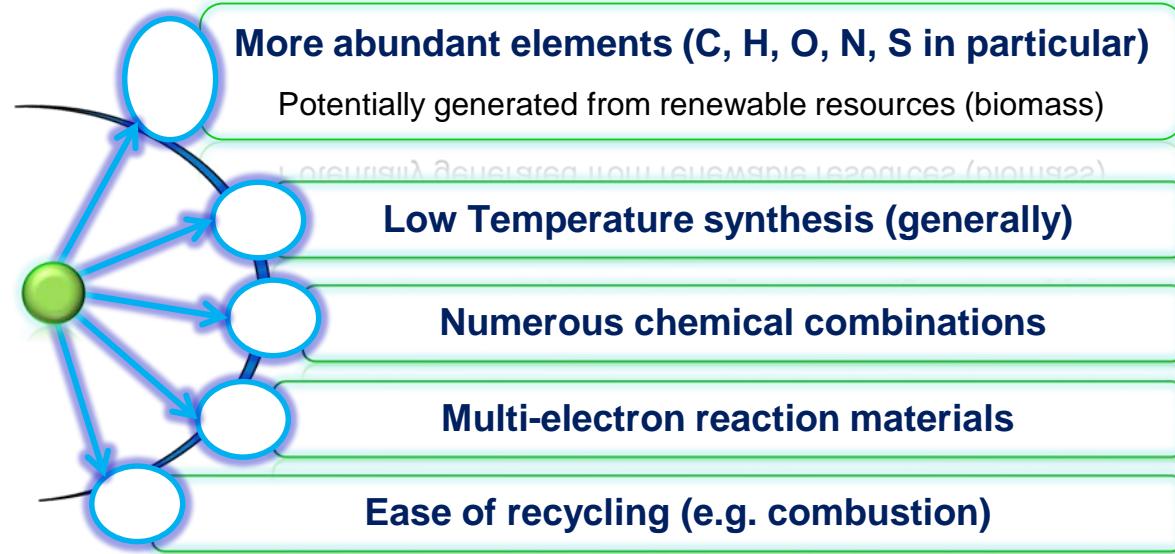
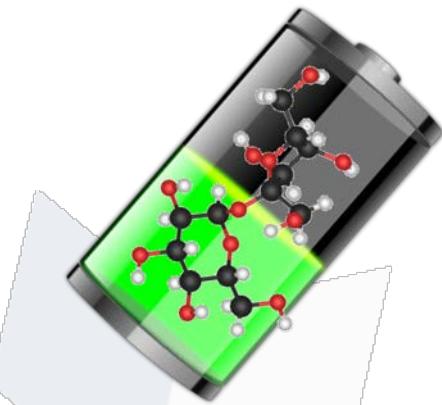
Dense crystal packing
(iono-covalent bonding)
1 atom = 1 redox center,
mostly

Less dense crystal packing
(van der Walls bonding)
several atoms = 1 redox
center, mostly

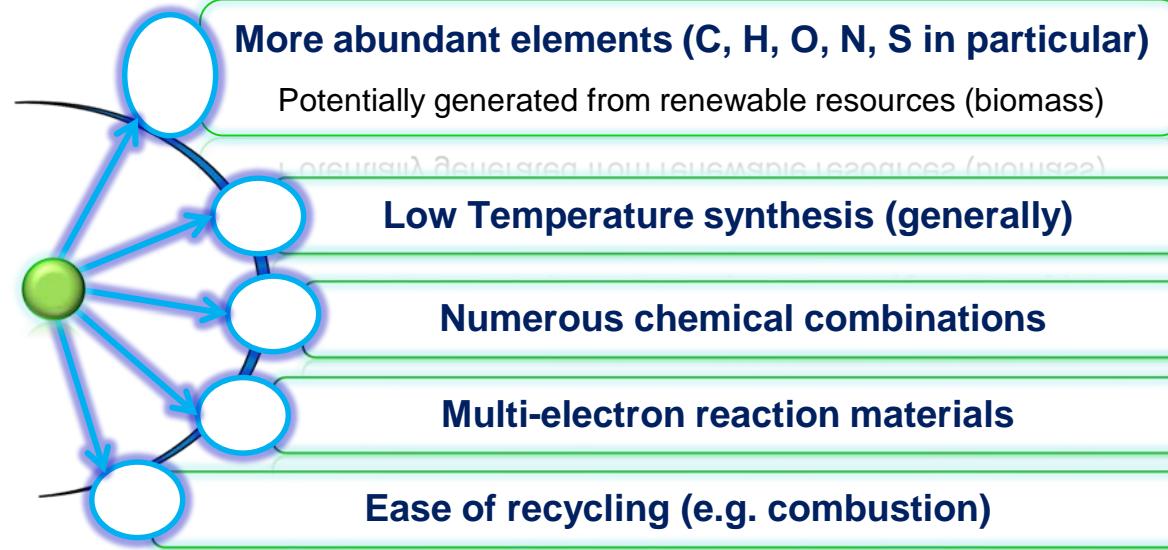
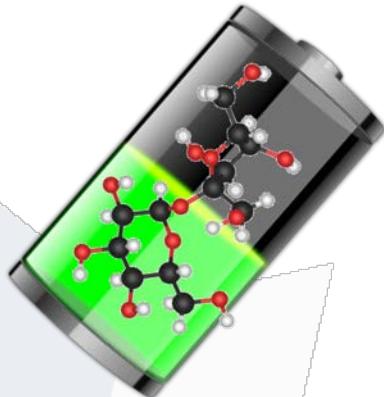
Organic electrode materials, CONS



Organic electrode materials, PROS



Organic electrode materials, PROS

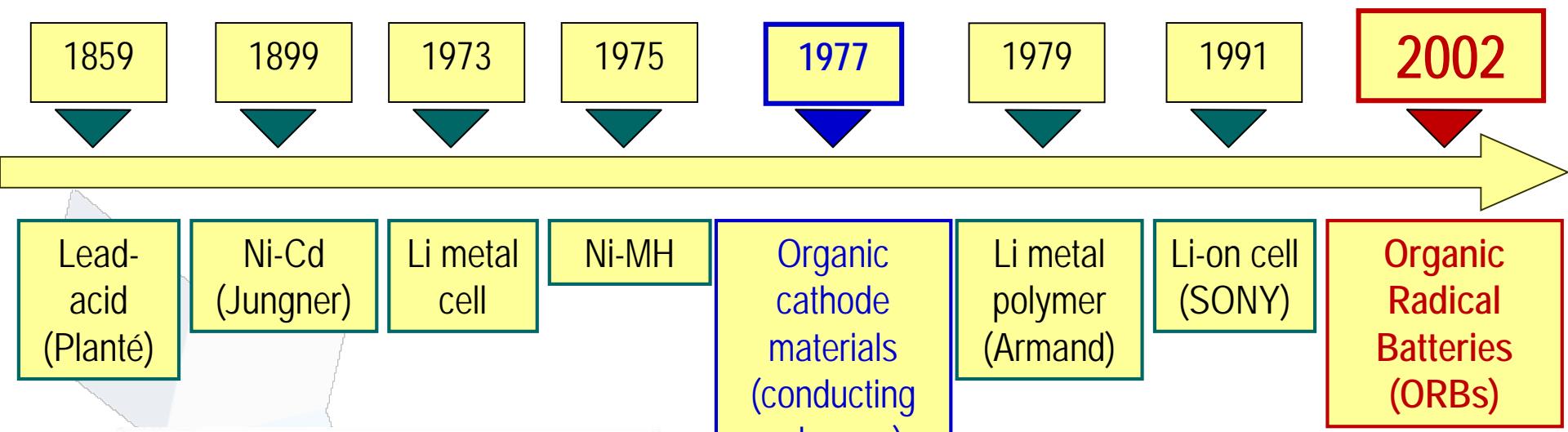


DIFFERENT CELL CONFIGURATIONS:

**Li-ion, Na-ion, Dual-ion cell,... in Aqueous or Non-aqueous electrolyte
with soluble redox species (RFB) or solid electrode materials**



... promoting organic batteries?



The Nobel Prize in Chemistry 2000



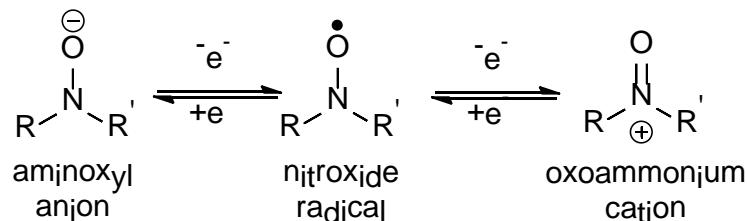
Alan J. Heeger
Prize share: 1/3



Alan G. MacDiarmid
Prize share: 1/3



Hideki Shirakawa
Prize share: 1/3



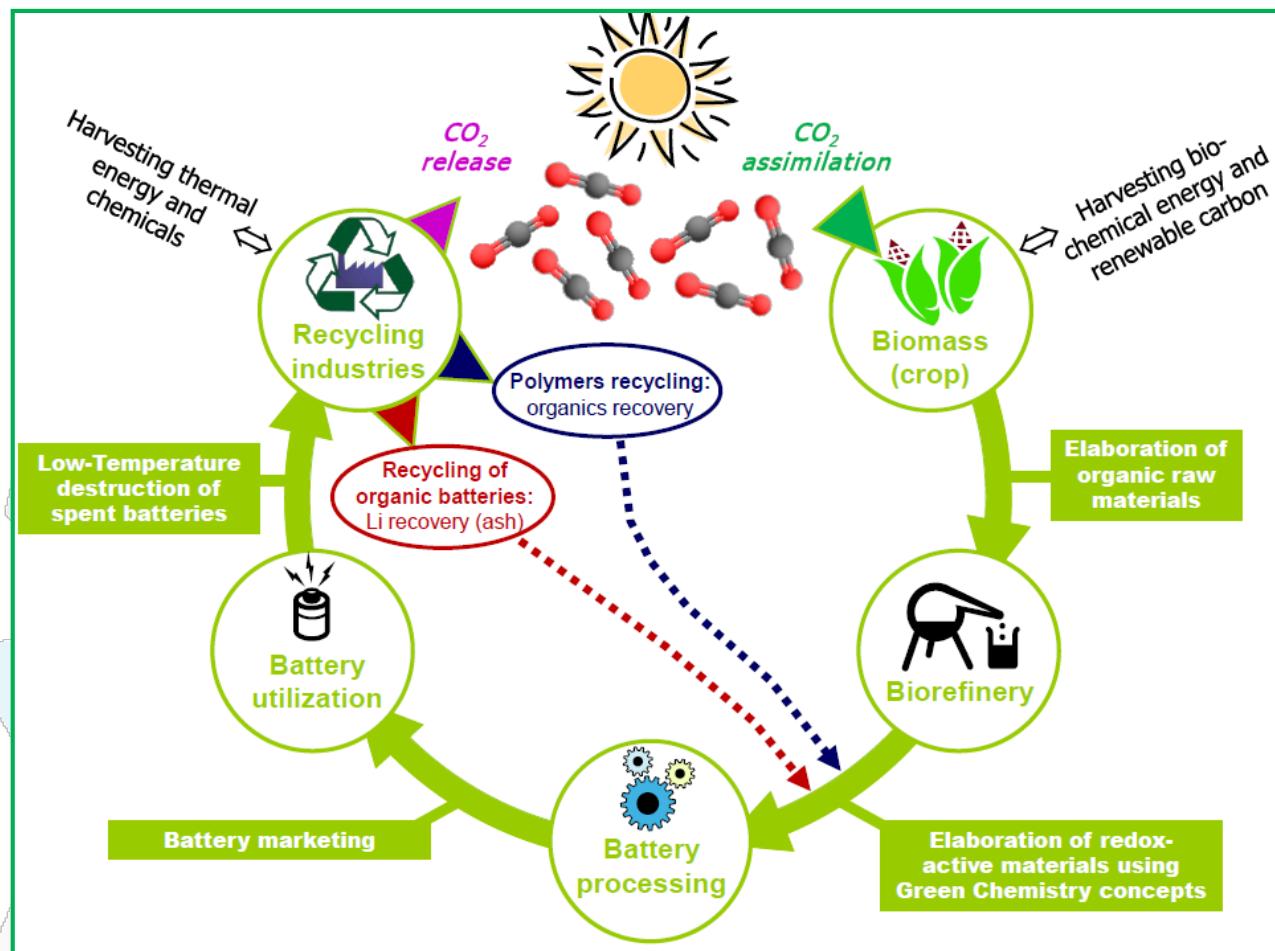
1977

1979

1991

2002

2008



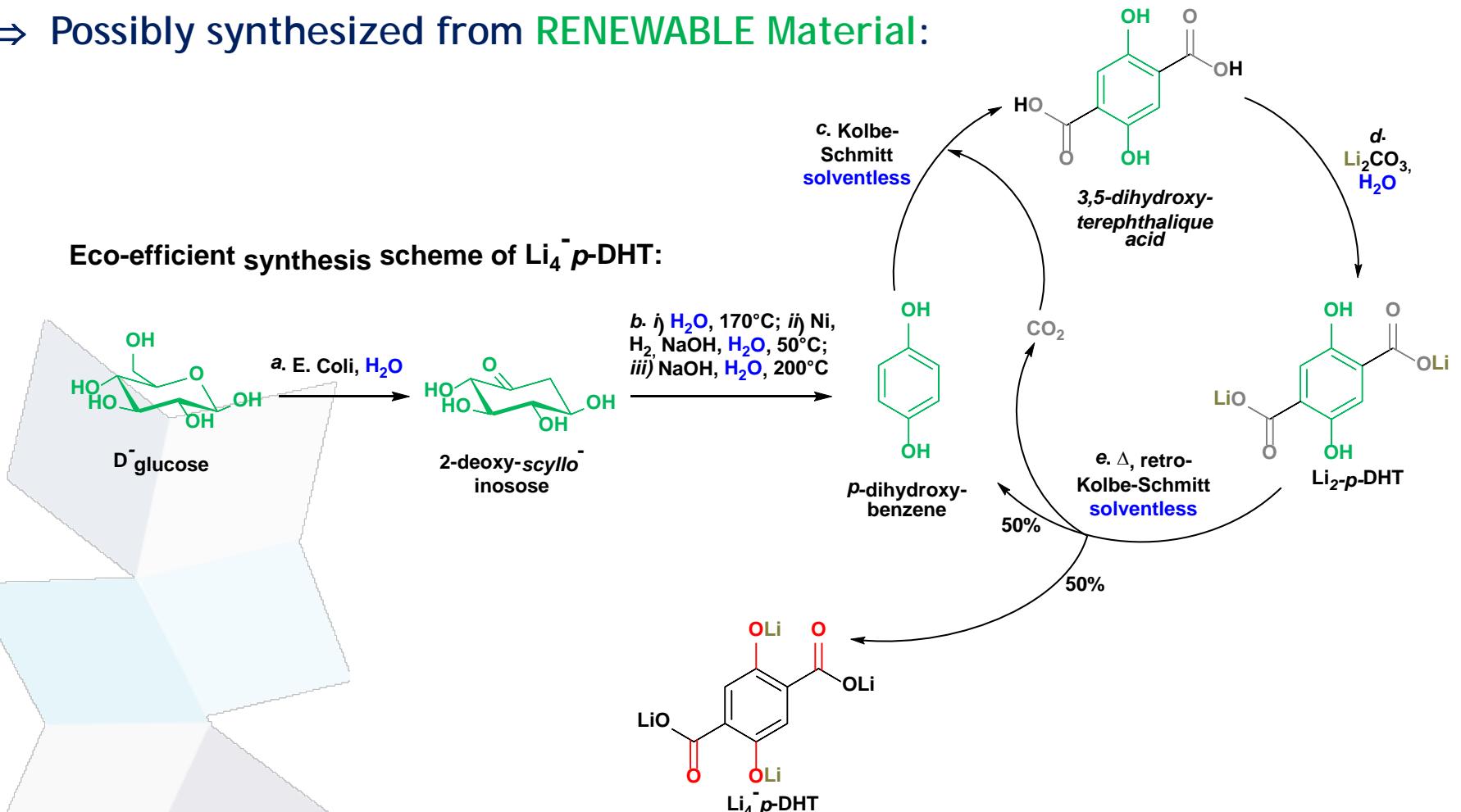
Virtuous circle

ChemSusChem, 2008
Energy Environ. Sci., 2011

n-type Li cathode: case of $\text{Li}_4\text{-p-DHT}$

Dilithium (2,5-dilithium-oxy)-terephthalate

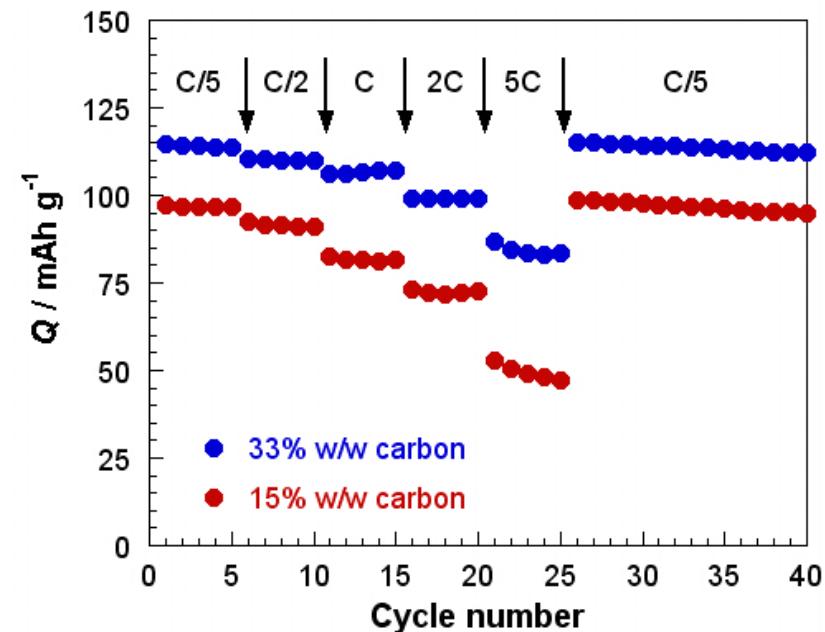
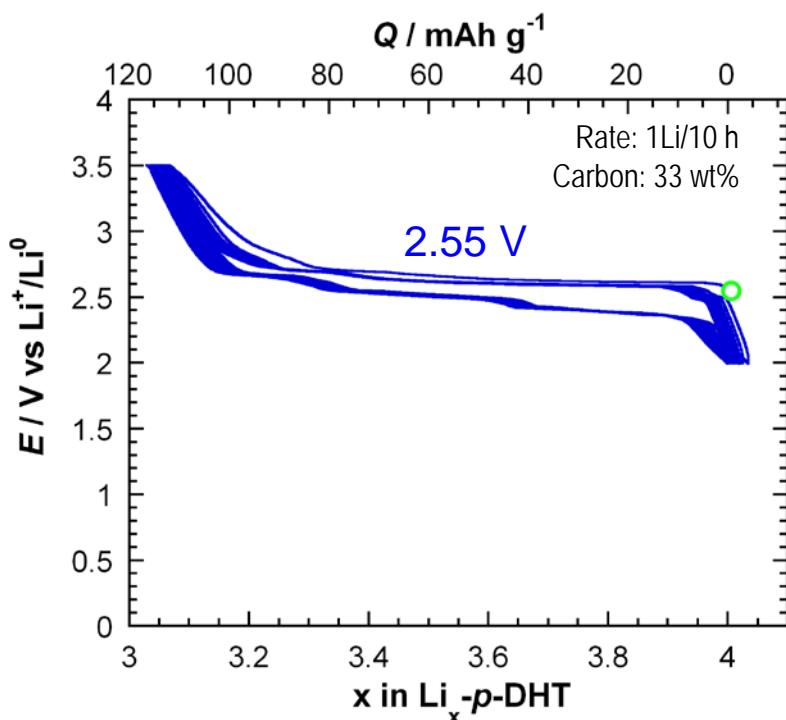
⇒ Possibly synthesized from RENEWABLE Material:



□ *Energy Environ. Sci.*, **6**, 2124 (2013).

n-type Li cathode: case of Li_4 -p-DHT

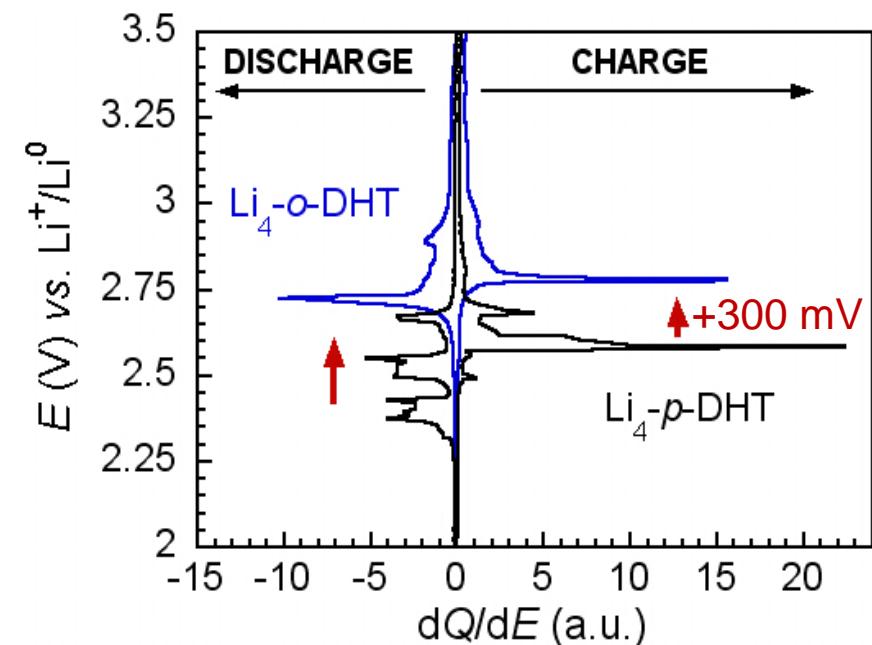
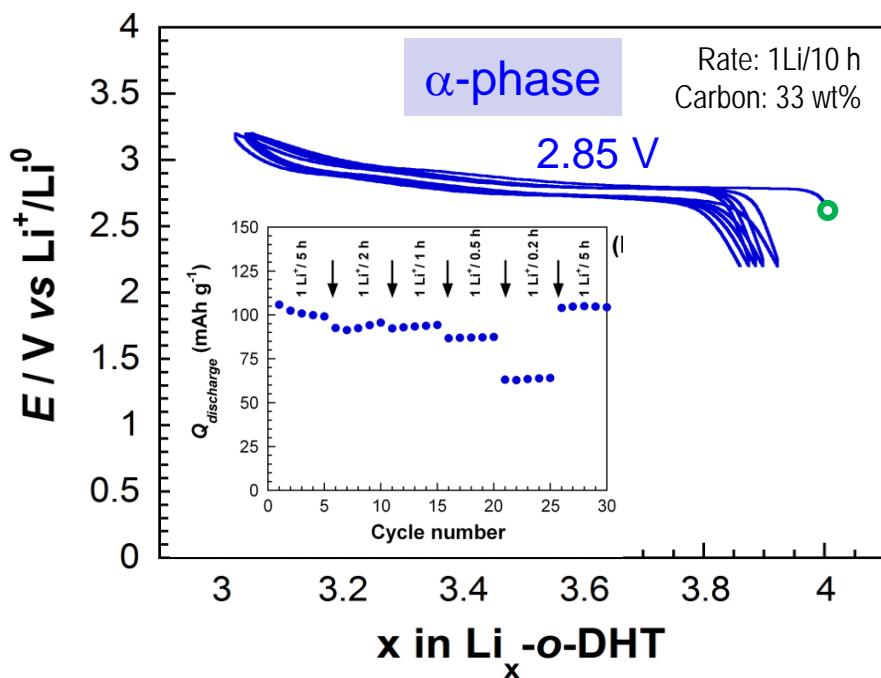
Dilithium (2,5-dilithium-oxy)-terephthalate



- Good electrochemical performance upon cycling including at high rate with no optimization (no binder, no complex formulation)

n-type Li cathode: case of $\text{Li}_4\text{-o-DHT}$

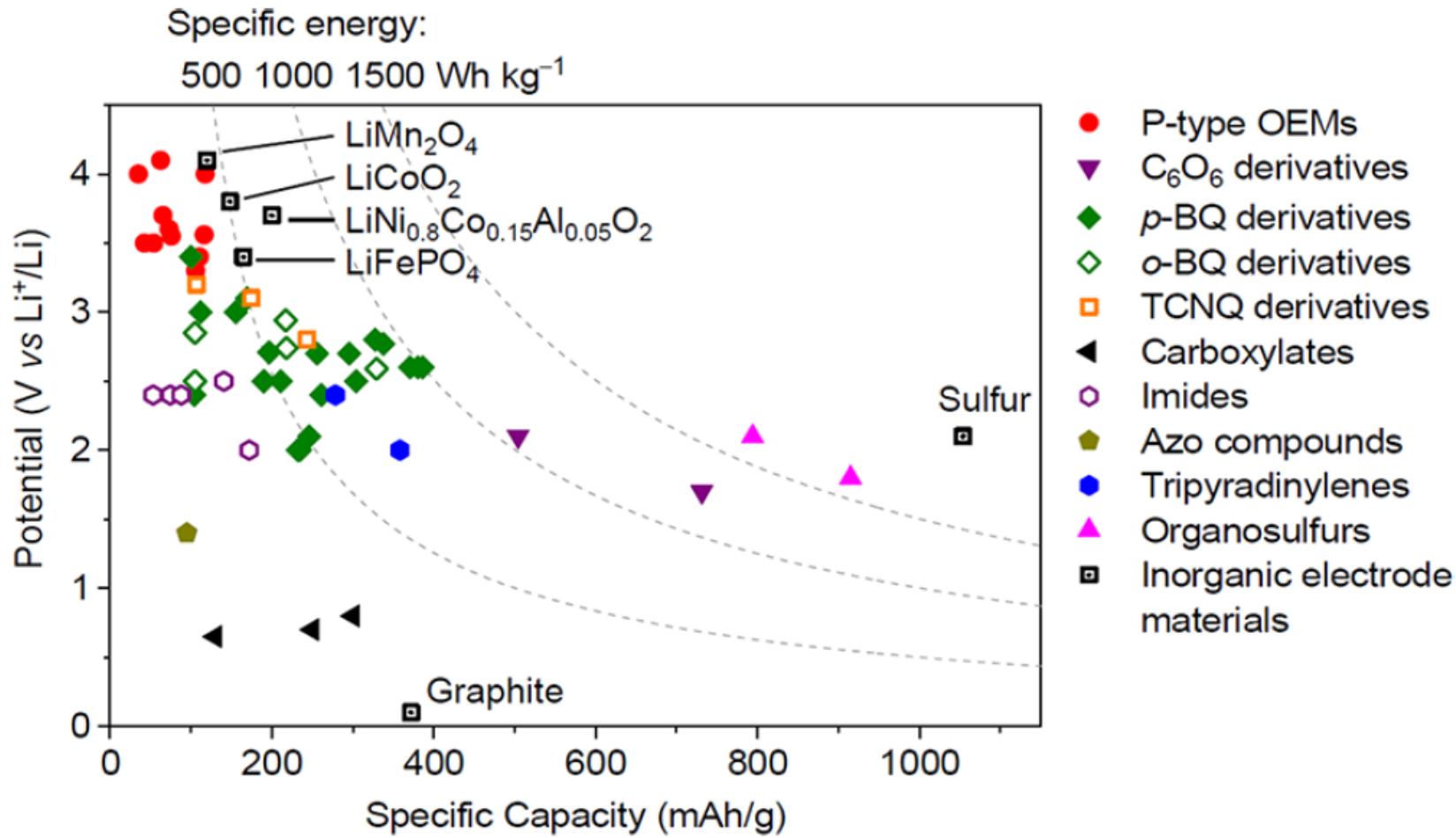
Dilithium (2,3-dilithium-oxy)-terephthalate



→ Voltage gain at least of + 300 mV

□ ACS Appl. Mater. Interfaces, 6, 10870 (2014).

Comparison with the state-of-the-art Organic vs Inorganic electrode materials for Li batteries

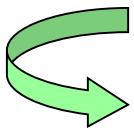


Summary



Organic materials can be considered as an alternative chemistry vs metals and other inorganic compounds

⇒ Richness of Organic structures and beyond!



Developing anion-ion batteries (potentially free of metals = molecular batteries) is possible

⇒ Proof of concept for an anion-ion battery made of crystalized organic materials



This research field is still in its infancy, and in terms of electrochemical performance, we cannot compete with regular Li-ion cells

PERFORMANCE vs ABUNDANCE?

CHEMICAL REVIEWS

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Review

Opportunities and Challenges for Organic Electrodes in Electrochemical Energy Storage

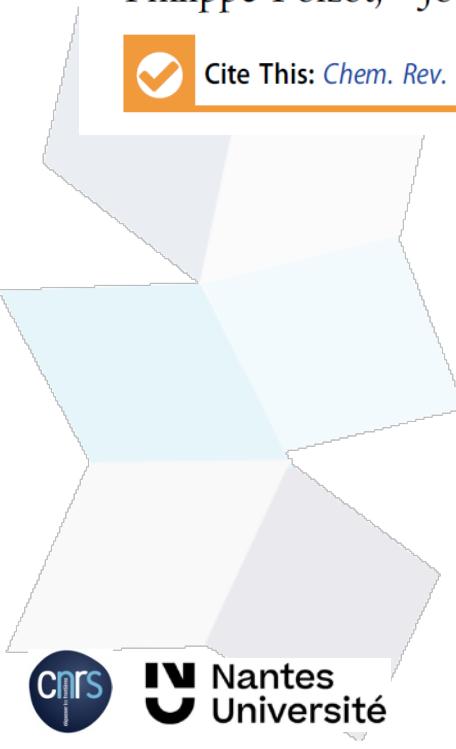
Philippe Poizot,* Joël Gaubicher, Stéven Renault, Lionel Dubois, Yanliang Liang, and Yan Yao



Cite This: *Chem. Rev.* 2020, 120, 6490–6557



Read Online



Thank you for your attention



Selected contributions // Organic Batteries & Perspective:

- *Clean energy new deal for a sustainable world: from non-CO₂ generating energy sources to greener electrochemical storage devices*, *Energy Environ. Sci.*, **4**, 2003 (2011)
- *Progress in all-organic rechargeable batteries using cationic and anionic configurations: Toward low-cost and greener storage solutions?*, *Curr. Op. in Electrochem.*, **9**, 70 (2018)
- *Opportunities and Challenges for Organic Electrodes in Electrochemical Energy Storage*, *Chem. Rev.*, **120**, 6490–6557 (2020)
- *A perspective on organic electrode materials and technologies for next generation batteries*, *J. Power Sources*, **482**, 228814 (2021).
- *2021 roadmap for sodium-ion batteries*, *J. Phys-Energy*, **3**, 031503 (2021).